

TYPES OF STEAM

Engineering Thermodynamics & Fluid Systems

A Comprehensive Visual Guide for Undergraduate Engineering Students

What Is Steam? — The Fundamentals

Steam is water (H_2O) in its vapor phase — produced when liquid water absorbs enough thermal energy to undergo phase transition above $100^\circ C$ at 1 atm pressure.

Formula: H_2O — same molecule, different energy state

Boiling Point: $100^\circ C$ at 1 atm (sea level)

Latent Heat: 2260 kJ/kg (vaporization energy)

Specific Volume: $1.67 \text{ m}^3/\text{kg}$ at atmospheric pressure

PHASE TRANSITION DIAGRAM



Why Steam Matters — Global Significance



80%

of electricity
globally via steam turbines

2260

kJ/kg
latent heat of
vaporization

700+

°C
max steam temp in
advanced plants

\$180B

global steam
market size (2023)

KEY INDUSTRIES POWERED BY STEAM

Power Generation

Rankine cycle turbines

Food Processing

Sterilization & cooking

Petrochemicals

Distillation & cracking

Textiles

Dyeing & finishing

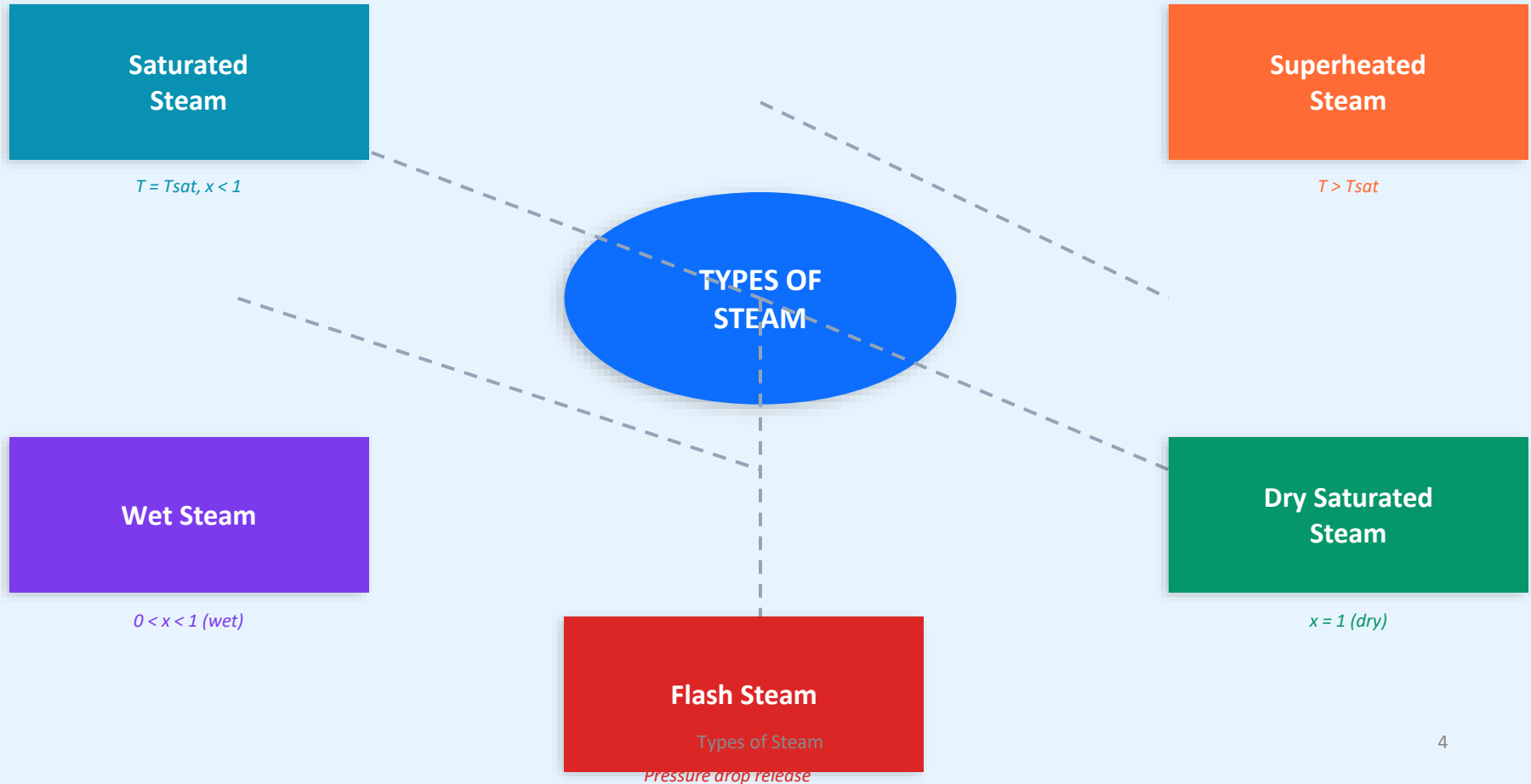
Paper & Pulp

Drying processes

HVAC Systems

District heating

Classification of Steam — Mind Map Overview



Wet Steam — Liquid-Vapor Mixture

Wet steam contains both water droplets and steam vapor coexisting in thermodynamic equilibrium. Dryness fraction (x) defines the quality.

DRYNESS FRACTION (x) SCALE



KEY FORMULA

$$x = \frac{m_{\text{steam}}}{m_{\text{steam}} + m_{\text{water}}}$$

$$h = h_f + x \cdot h_{fg} \quad (\text{Enthalpy of wet steam})$$

Temperature: Equal to saturation temp

Pressure: Saturation pressure

Dryness (x): $0 < x < 1$

Appearance: Foggy, misty, opaque

Efficiency: Low — droplets cause erosion

Applications: Low-grade heating, rare in turbines

Dry Saturated Steam — The Ideal Boundary State

Dryness

$$x = 1.0$$

State

Saturated
Vapor

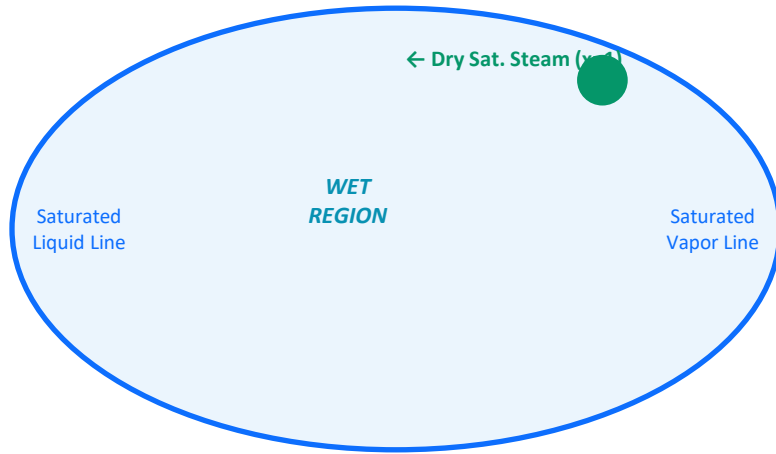
Temperature

$$= T_{\text{sat}}$$

Visibility

Transparent
Colorless

T-s SATURATION DIAGRAM



⚡ **High Energy Density**

Contains max latent heat, ideal for heat transfer

⚖ **Equilibrium State**

Boundary between wet steam & superheated steam

⚙ **Turbine Entry**

Common inlet condition for steam turbines

📖 **Steam Tables**

Reference state for thermodynamic property lookup

💧 **Critical Point**

Ceases to exist above 374.14°C (critical temperature)

Superheated Steam — Beyond Saturation

Superheated steam is heated beyond its saturation temperature at constant pressure. It behaves as an ideal gas — no liquid phase present, higher enthalpy, higher work output.

DEGREES OF SUPERHEAT

Low
<100°C superheat

Medium
100–250°C

High
250–400°C

Ultra-High
>400°C

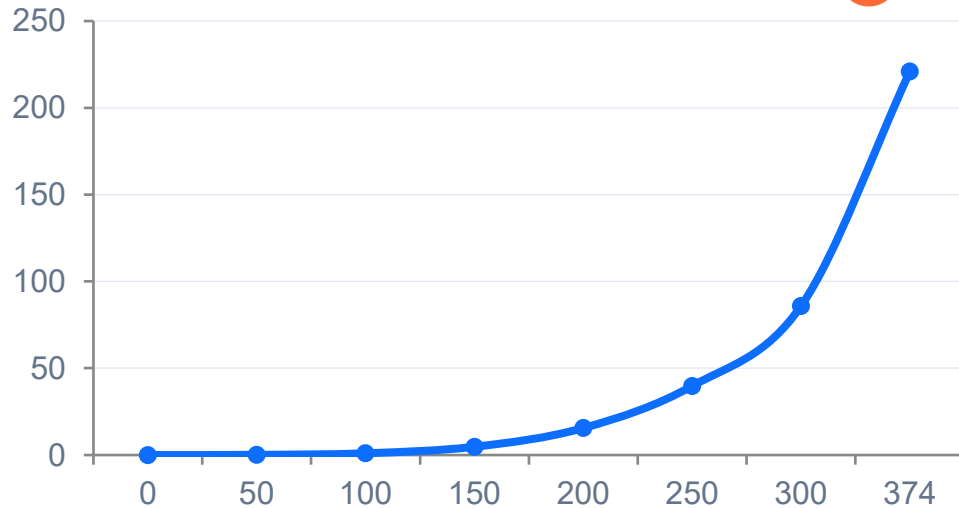
Property	Sat. Steam	Superheated	Advantage
Temperature	Tsat	Tsat + ΔT	Higher
Enthalpy	hg	hg + Cp $\cdot\Delta T$	More Work
Specific Vol.	vg	> vg	More Flow
Visibility	May fog	Invisible	Quality
Efficiency	Moderate	High	Better

Types of Steam  Superheat above 600°C requires special alloy steels to handle oxidation and creep.

Saturated Steam — The Equilibrium State

Saturation Pressure (bar) vs
Temperature (°C)

Critical Point
374°C / 220.9 bar



STEAM TABLE REFERENCE POINTS

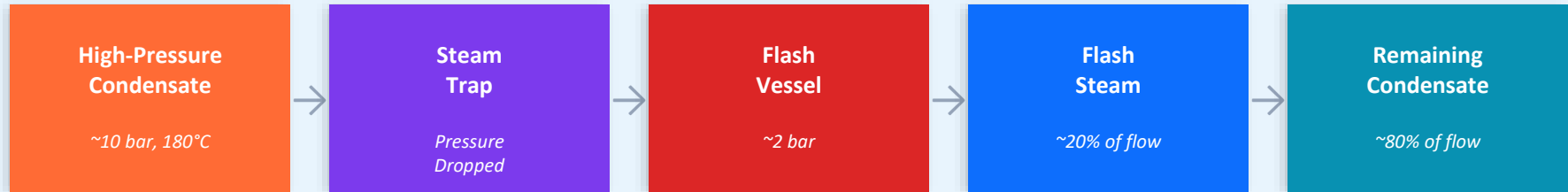
T (°C)	P (bar)	hf (kJ/kg)	hg (kJ/kg)
50	0.123	209.3	2592
100	1.013	419.1	2676
150	4.758	632.2	2746
200	15.54	852.4	2793
250	39.76	1085.4	2801

💡 At saturation: adding heat → superheated steam (constant P) | removing heat → wet steam. The saturation line is a thermodynamic boundary!

Flash Steam — Energy Recovery from Condensate

Flash steam forms when high-pressure hot condensate is released to lower pressure — excess enthalpy instantly vaporizes a portion of the liquid.

FLASH STEAM GENERATION PROCESS



⚡ FLASH STEAM FRACTION

$$\% \text{ Flash} = (hf1 - hf2) / hfg2 \times 100$$

Where:

hf1 = enthalpy at high pressure

hf2 = enthalpy at low pressure

hfg2 = latent heat at low pressure



Industrial Plants

Recover waste condensate energy



Heat Exchangers

Secondary heating with flash steam

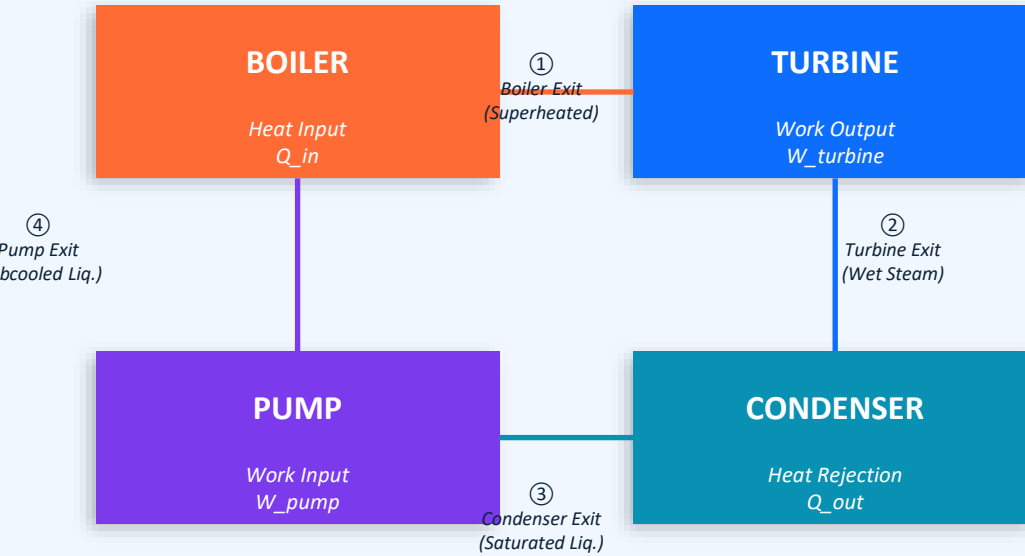


Energy Savings

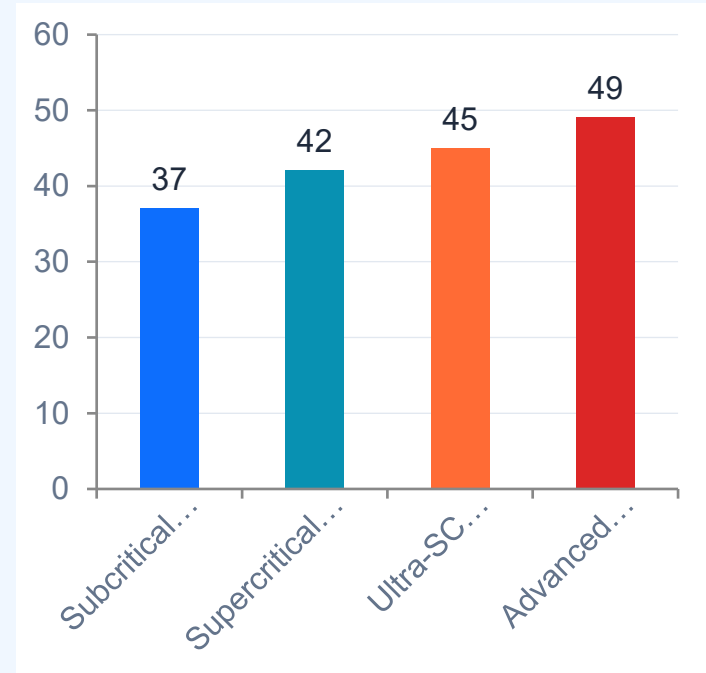
Up to 10-20% energy recovery possible

Steam in the Rankine Cycle — Power Generation

THE BASIC RANKINE CYCLE



POWER PLANT EFFICIENCY BY STEAM TYPE



Comparative Properties — All Steam Types at a Glance

Property	Wet Steam	Dry Saturated	Superheated	Flash Steam
Temperature	= T_{sat}	= T_{sat}	> T_{sat}	= T_{sat} (new P)
Dryness (x)	$0 < x < 1$	$x = 1$	N/A	~0.1–0.3
Enthalpy	$h_f + x \cdot h_{fg}$	h_g	$h_g + C_p \cdot \Delta T$	Low
Visibility	Misty/Cloudy	Transparent	Invisible	White fog
Pressure	Saturation P	Saturation P	Any (given)	New lower P
Industrial Use	Rarely preferred	Heat transfer	Turbines/Power	Energy recovery
Efficiency	Low	Moderate	High	Moderate

✦ Superheated steam offers maximum efficiency and turbine protection. Dry saturated is used for process heating. Wet steam should generally be avoided in machinery.

POWER GENERATION

Thermal Power Plants

Superheated steam at 540°C drives turbines generating 60–70% of global electricity

Nuclear Plants

Saturated/slightly superheated steam from reactor heat drives turbines

Geothermal

Natural dry or wet steam from earth drives turbines directly

Combined Cycle

Waste heat from gas turbines creates steam for secondary power generation

INDUSTRIAL PROCESSES

Food & Beverage

Autoclaving, pasteurization, sterilization at 121°C (saturated steam)

Pharmaceutical

GMP-grade clean steam for sterilization of vessels and instruments

Pulp & Paper

Drying cylinders use saturated steam; dryer sections run at ~160°C

Textile Industry

Steam pressing, dyeing, and fabric relaxation processes

District Heating

Saturated/hot water distribution for residential and commercial HVAC

Advantages vs. Challenges — Engineering Trade-offs



✓ ADVANTAGES

✓ High Energy Density

2260 kJ/kg latent heat enables compact heat exchangers

✓ Precise Temp Control

Saturation T-P relationship gives predictable temperatures

✓ Clean Working Fluid

Water is non-toxic, abundant, and recyclable

✓ Scalable

From 1 kW domestic boilers to 1000+ MW power plants

✓ Versatile

Heating, sterilization, power, drying — one fluid, many uses

✓ Established Tech

200+ years of engineering knowledge and infrastructure

⚠ CHALLENGES

✗ Corrosion Risk

Oxygen and CO₂ in steam cause severe boiler/pipe corrosion

✗ Water Hammer

Condensate pockets create dangerous pressure surges

✗ High Maintenance

Boilers, traps, and valves require frequent inspection

✗ Energy Losses

Heat distribution losses can reach 20–30% in old systems

✗ Safety Hazards

High-pressure steam explosions are catastrophic

✗ Scale/Deposits

Dissolved minerals form scale, reducing heat transfer

STEAM QUALITY HIERARCHY

↑
QUALITY

Ultra-Pure / Clean Steam

>99.9% vapor, pharmaceutical grade

Superheated Steam

100% vapor + superheat, $x = \text{undefined}$

Dry Saturated ($x=1$)

All vapor, no liquid droplets

Wet Steam ($x=0.9$)

10% liquid by mass — undesirable

Very Wet ($x<0.8$)

Dangerous — erodes blades

QUALITY MEASUREMENT METHODS

Throttling Calorimeter

Steam throttled to lower pressure, then superheated. x calculated from enthalpy balance. *Range: $x = 0.85-1.0$*

Separating Calorimeter

Moisture mechanically separated. $x = m_{\text{steam}} / (m_{\text{steam}} + m_{\text{water}})$. *Range: $x = 0.5-0.9$*

Combined Calorimeter

Series combination of separating + throttling for very wet steam. *Range: $x < 0.8$*

Electrical Methods

Modern capacitance and microwave probes measure moisture content inline. *Range: Real-time*

STEAM TECHNOLOGY EVOLUTION TIMELINE



EMERGING STEAM TECHNOLOGIES

Supercritical CO₂

sCO₂ cycles replacing steam in some applications — higher density, smaller turbines

Green Hydrogen Steam

Steam from electrolysis-based processes using renewable energy sources

Solar Steam

Concentrated solar power generating superheated steam at 550°C+ without fossil fuels

AI Steam Management

Machine learning optimizing steam distribution, trap monitoring, and energy recovery


700°C Ultra-SC

Nickel alloy turbines enabling 50%+ efficiency — EU COMTES700 project

Plasma-Assisted Steam

Non-thermal plasma treatment improving steam quality for pharmaceutical applications

TYPES OF STEAM — SUMMARY

 WET STEAM	 DRY SAT.	 SUPERHEATED	 FLASH STEAM	 SATURATED
Quality:	Quality:	Quality:	Quality:	Quality:
$0 < x < 1$	$x = 1$	$T > T_{sat}$	From P ↓	$T = T_{sat}(P)$
Char.:	Char.:	Char.:	Char.:	Char.:
Liquid + vapor mixture	Pure vapor at sat. conditions	Vapor above sat. temperature	Condensate at new low pressure	At saturation boundary
Used for:	Used for:	Used for:	Used for:	Used for:
Low-grade heating only	Process heat, heat exchangers	Turbines, high-efficiency power	Energy recovery	Broad industrial applications
Notes:	Notes:	Notes:	Notes:	Notes:
 Avoid in machinery	 Good for steady heating	 Best for power gen.	 Free energy recovery	 Easy to control

KEY TAKEAWAYS

01

Steam classification is based on dryness fraction (x), temperature relative to saturation, and pressure conditions

02

Superheated steam maximizes turbine efficiency; wet steam causes blade erosion — always superheat before turbine entry

03

The Rankine cycle is the backbone of 80% of global electricity generation, all powered by steam

04

Flash steam is 'free energy' — proper condensate return systems can recover 10–20% waste energy

05

Future: 700°C ultra-supercritical and solar steam technologies will push thermal efficiency beyond 50%

"Steam is not just water vapor — it is the invisible engine driving modern civilization."

Types of Steam