



### Across

2. Process of determining the accuracy of a measuring instrument.
4. Point in a titration where the amount of titrant added is just enough to react with the analyte.
6. Point at which the reaction is complete and the indicator changes color.
7. Relationship between the relative quantities of reactants and products in a chemical reaction.
8. Laboratory tool used to transfer a precise volume of liquid.
9. Solution that resists changes in pH when small amounts of acid or base are added.

### Down

1. Ion or molecule that binds to a central metal atom to form a complex.
2. Chemical species formed by the combination of a metal ion and a ligand.
3. Substance that changes color to signal the endpoint of a titration.
5. Solution added to determine the concentration of another solution.