

# Antibiotics

## **PENICILLINS**

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# HISTORY OF ANTIBIOTICS

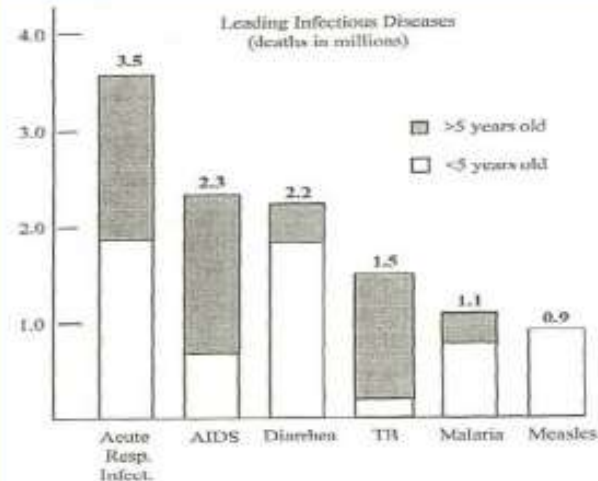
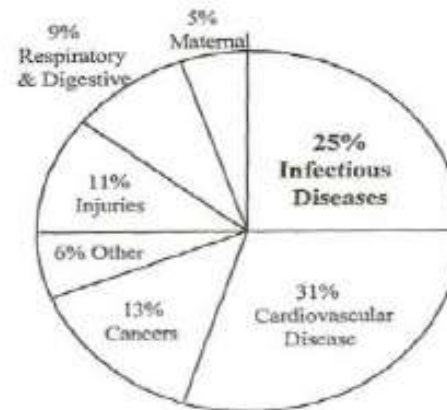
## The End of Infectious Disease

- From 1981 to 1995 deaths from infectious disease increased by 4.8% annually.
- In 1998 WHO estimated that over 13 million deaths worldwide were caused by infectious disease, almost a quarter of the total deaths in that period. That percentage was up to 26% in 2001.
- In 1995 the annual in-hospital costs associated with resistance of 6 bacterial species to a single antibiotic were estimated to be \$1.3 billion.
- 37 new human pathogens have been identified in the last 30 years.
- 12% of known human pathogens have been recognized as emerging or reemerging health threats

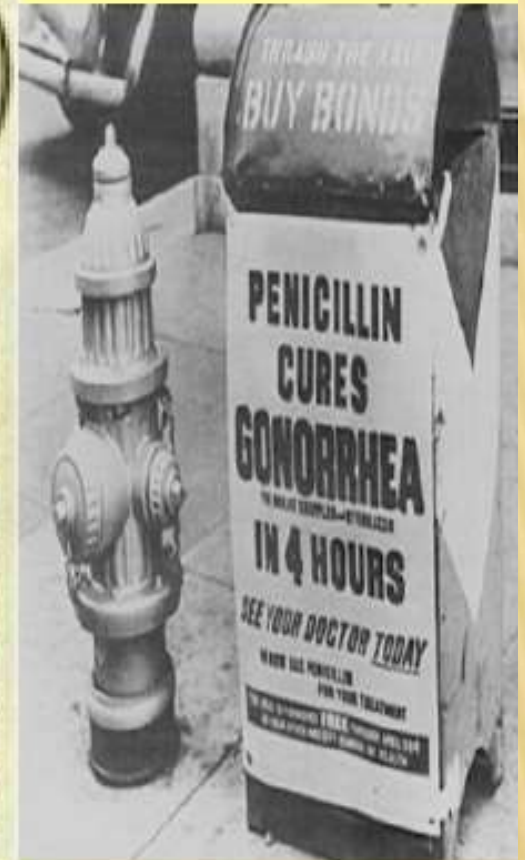
53,900,000 worldwide deaths in 1998

13,300,000 due to Infectious Disease

(does not include cardiovascular, respiratory or digestive diseases likely to have an infectious agent as source)



# HISTORY OF ANTIBIOTICS





# HISTORY OF ANTIBIOTICS

## Brief History of Antibiotics

- 1928- Penicillin discovered by Fleming
- 1932- Sulfonamide antimicrobial activity discovered {Erlisch}•
- 1943- Drug companies begin mass production of penicillin
- 1948- Cephalosporins precursor sent to Oxford for synthesis
- 1952- Erythromycin derived from *Streptomyces erythraeus*
- 1956- Vancomycin introduced for penicillin resistant staphylococcus
- 1962- Quinolone antibiotics first discovered
- 1970s- Linezolid discovered but not pursued
- 1980s- Fluorinated Quinolones introduced, making them clinically useful
- 2000- Linezolid introduced into clinical practice

# HISTORY OF ANTIBIOTICS

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## The History of Medicine

*2000 B.C.—Here, eat this root.*

*1000 A.D.—That root is heathen. Here, say this prayer.*

*1850 A.D.—That prayer is superstition. Here, drink this potion.*

*1920 A.D.—That potion is snake oil. Here, swallow this pill.*

*1945 A.D.—That pill is ineffective. Here, take this penicillin.*

*1955 A.D.—Oops...bugs mutated. Here, take this tetracycline.*

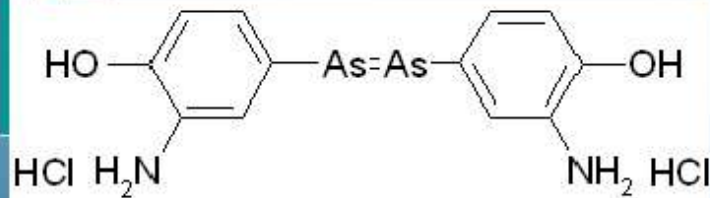
*1960–1999—39 more “oops.” Here, take this more powerful antibiotic.*

*2000 A.D.—The bugs have won! Here, eat this root.*

*—Anonymous (WHO, 2000)*

# HISTORY OF ANTIBIOTICS

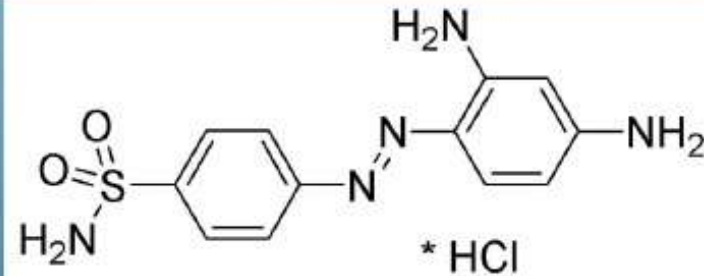
## Ehrlich's Magic Bullets





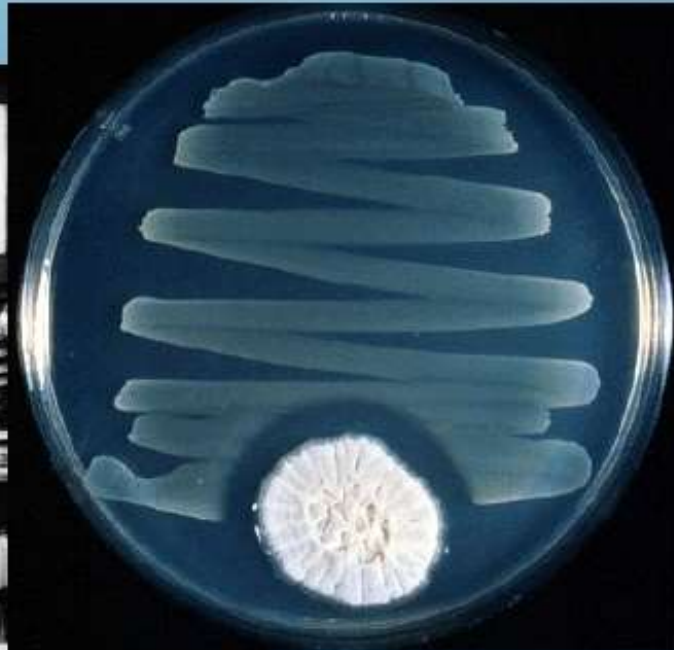
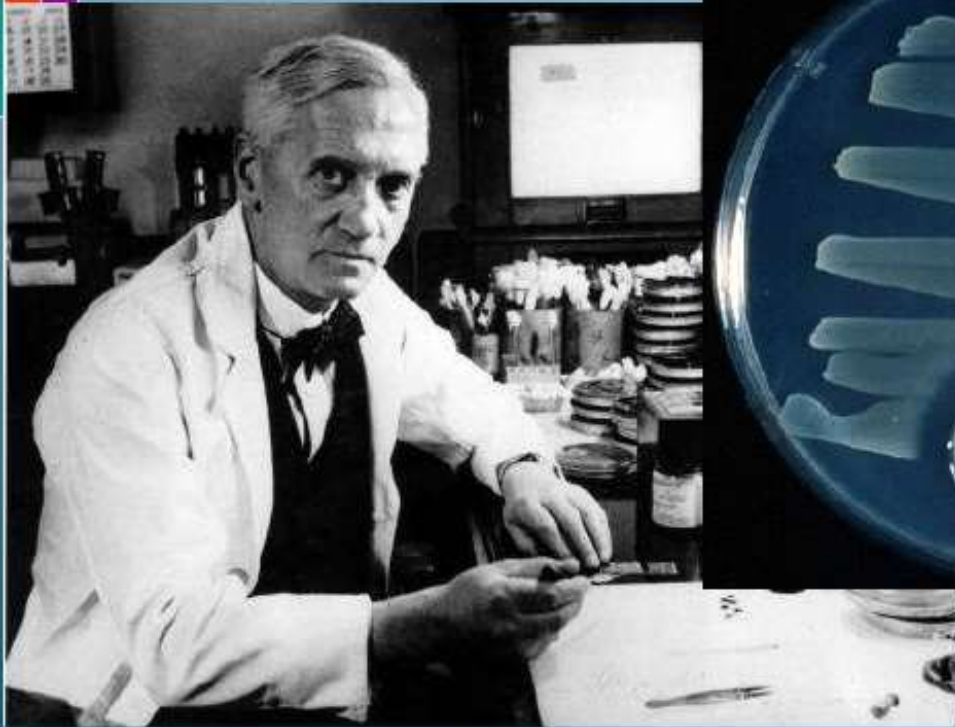
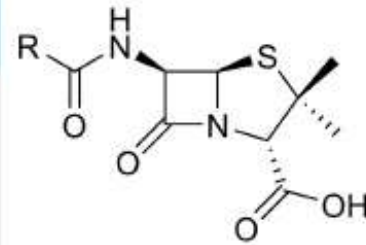
# HISTORY OF ANTIBIOTICS

## Gerhard Domagk - Prontosil



# HISTORY OF ANTIBIOTICS

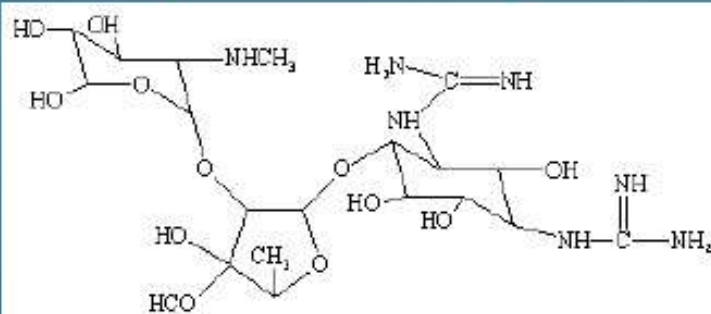
## Fleming and Penicillin





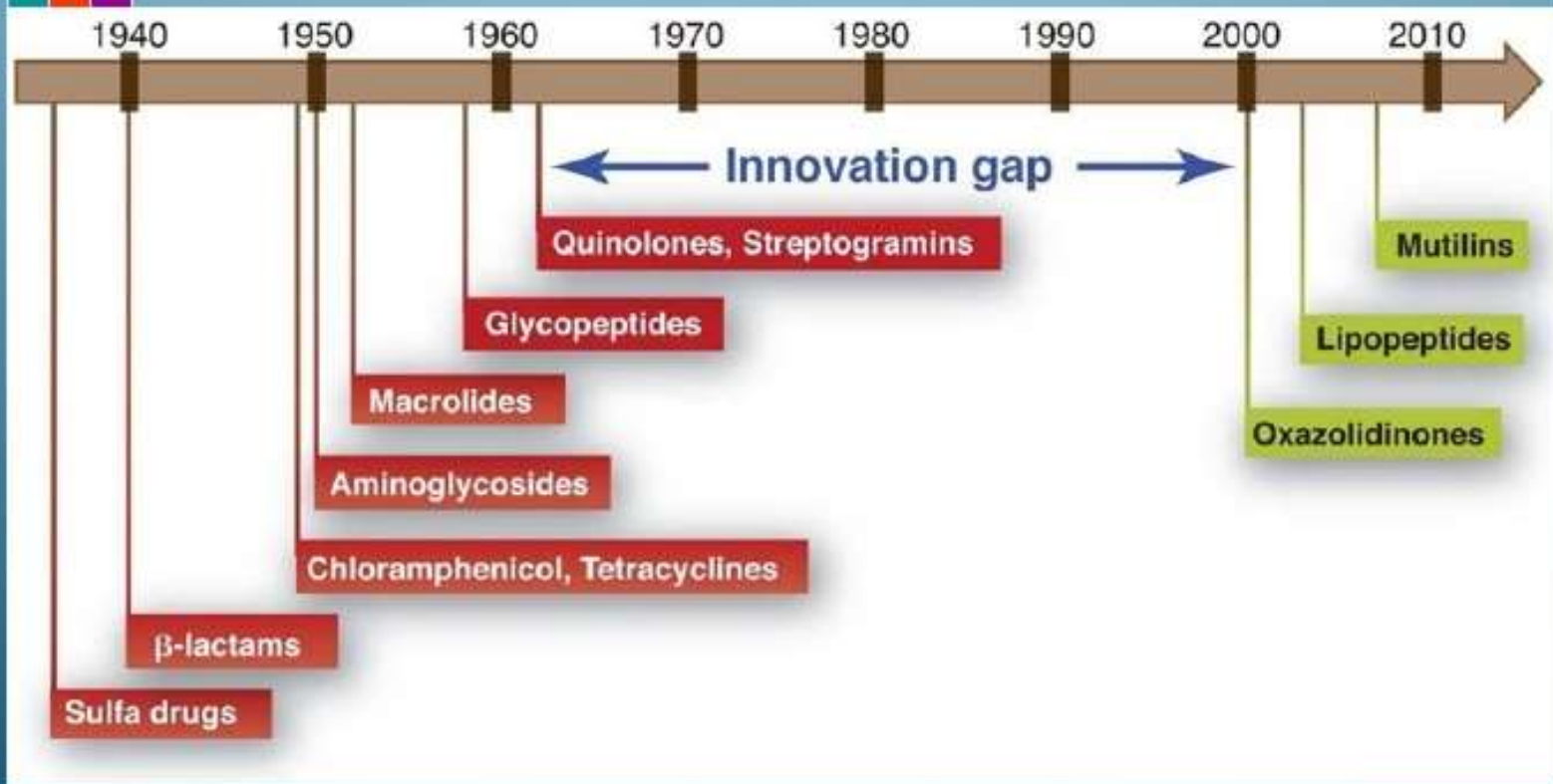
# HISTORY OF ANTIBIOTICS

Selman Waksman



# HISTORY OF ANTIBIOTICS

Between 1962 and 2000, no major classes of antibiotics were introduced

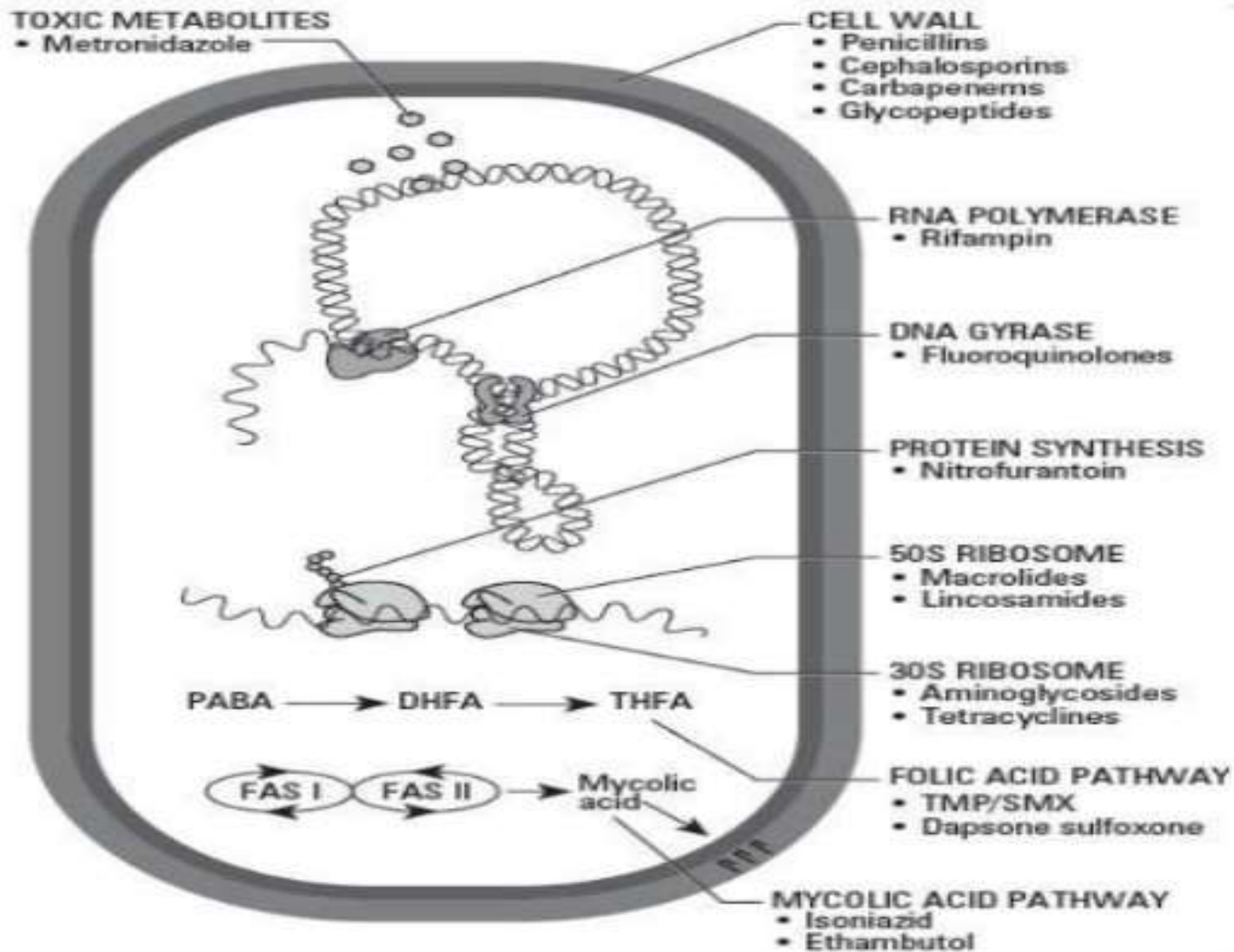


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# PENCILLINS



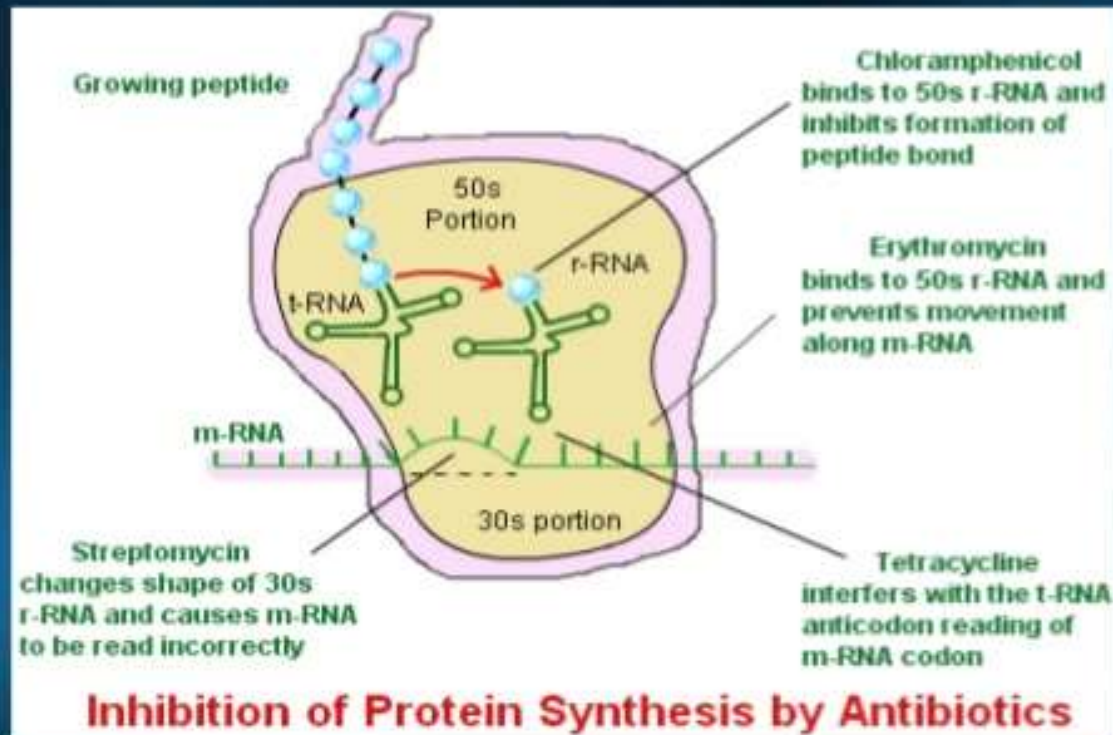
# MECHANISM OF ACTION



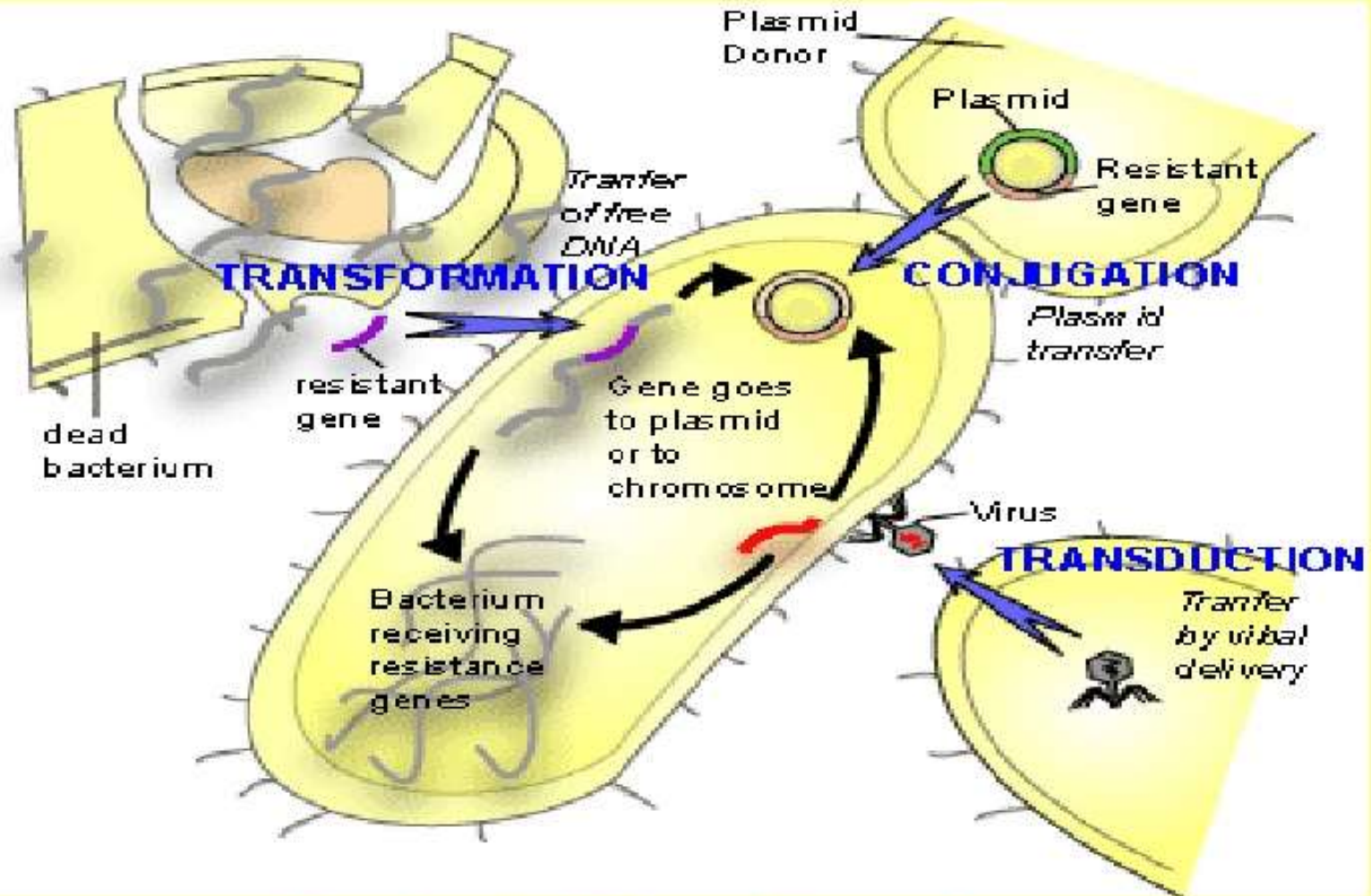
. Mechanism of action of antibiotics

# MECHANISM OF ACTION

## *Inhibition of protein synthesis*



# MECHANISM OF ACTION

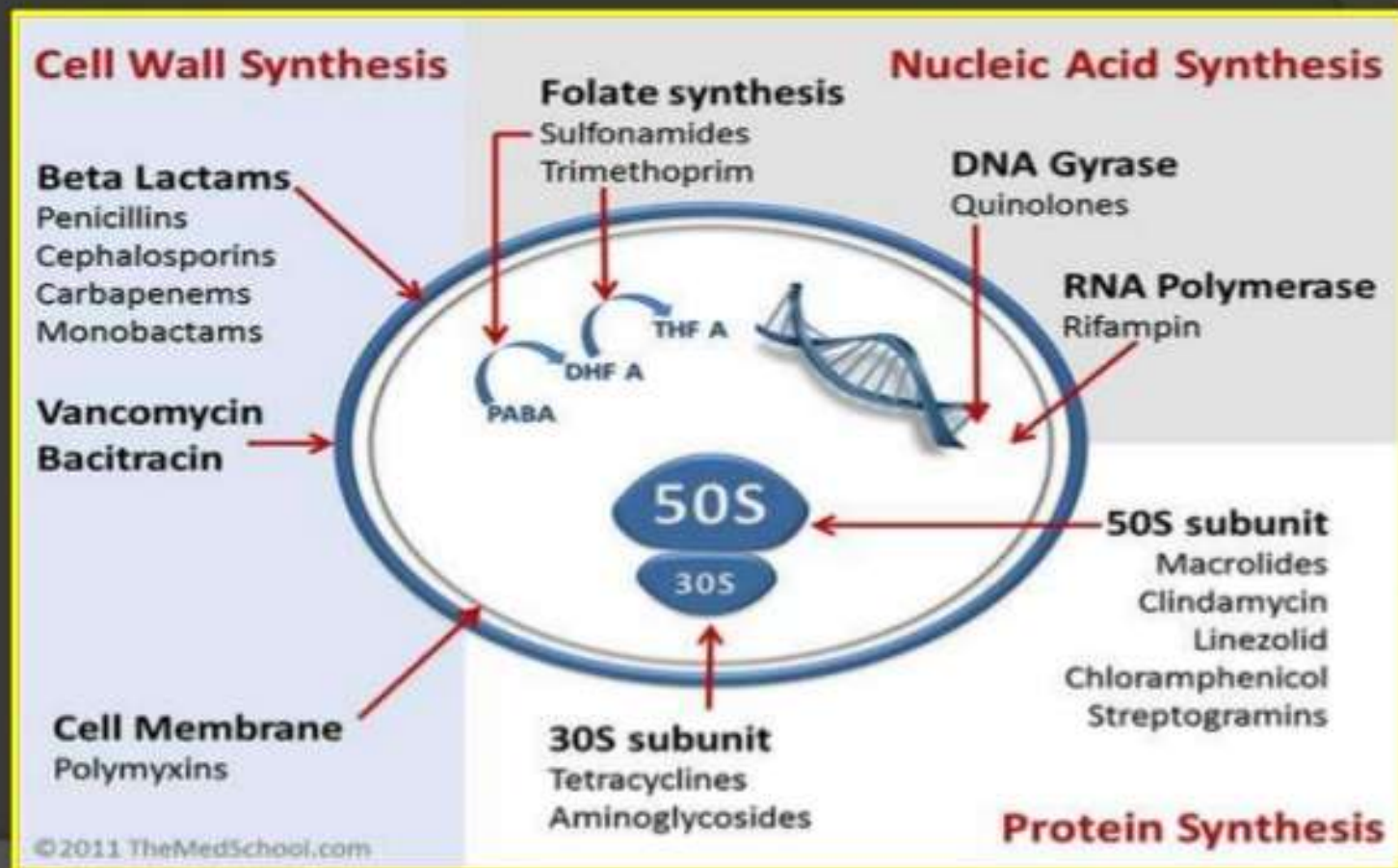


Horizontal Gene Transfer



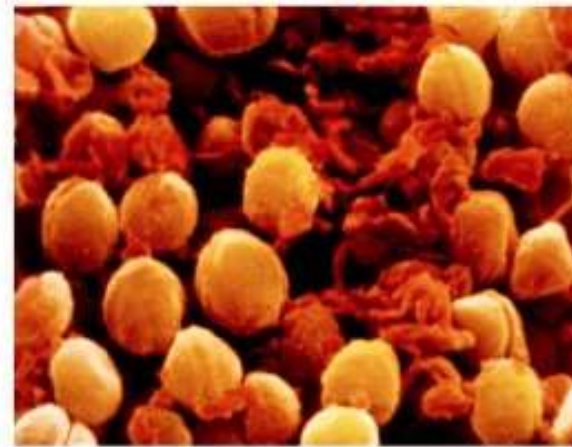
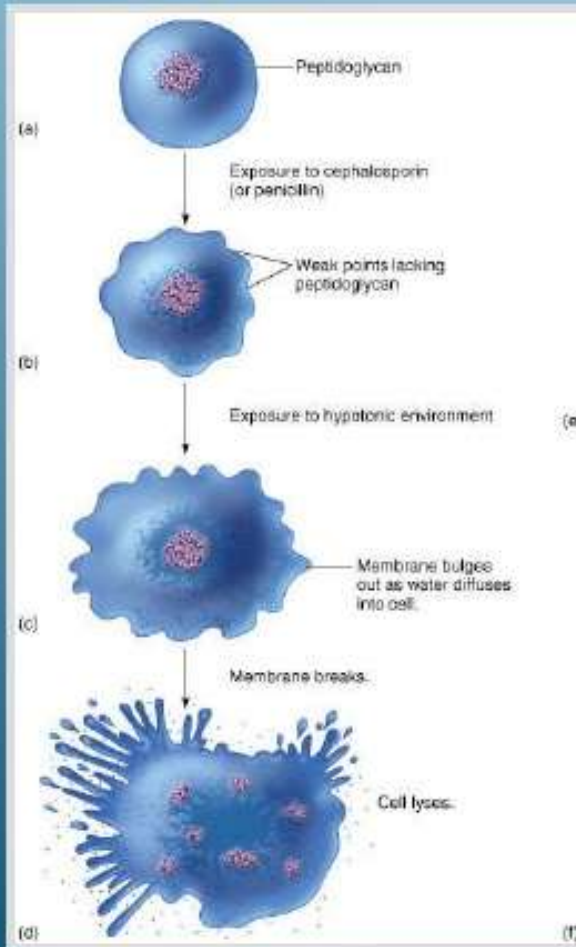
# MECHANISM OF ACTION

*On the basis of mechanism of action:*

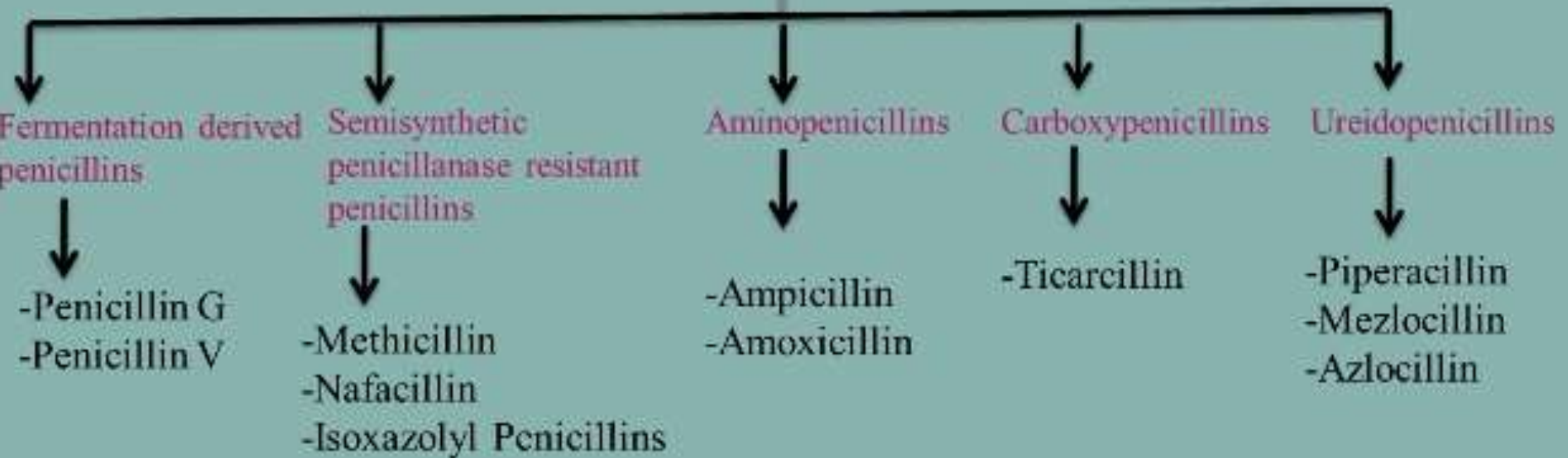


# MECHANISM OF ACTION

Antibiotics weaken the cell wall, and cause the cell to lyse



## Classification of Penicillins



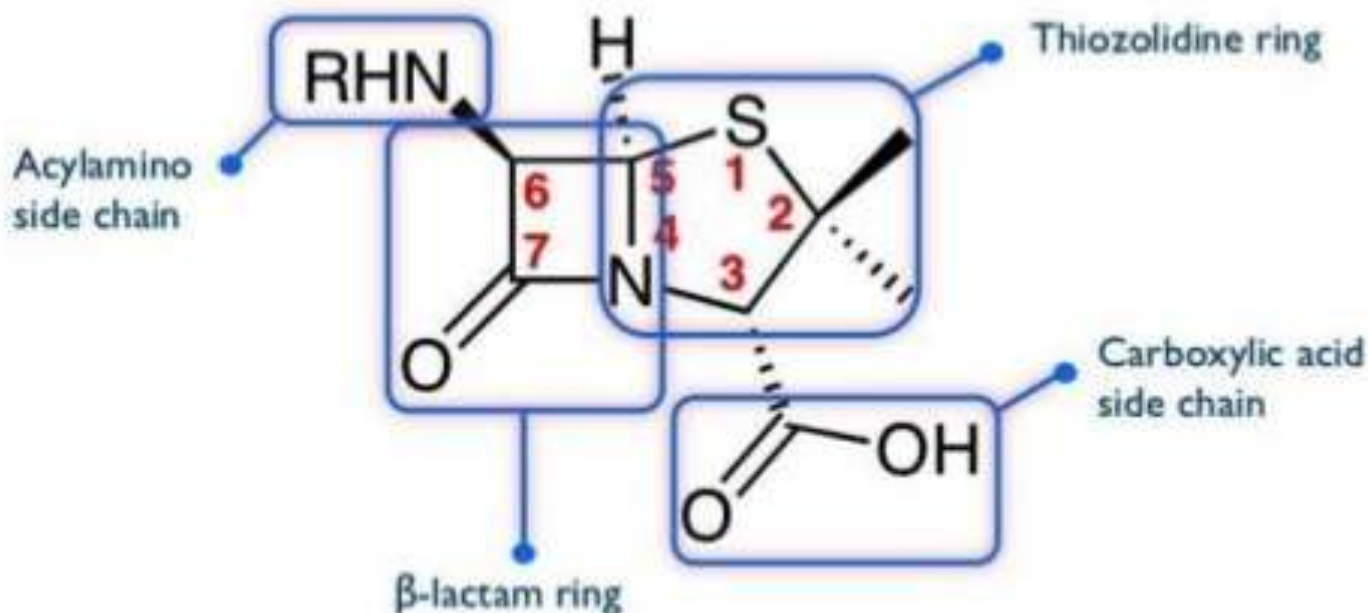


# CHEMISTRY OF PENICILLINS

## STRUCTURE

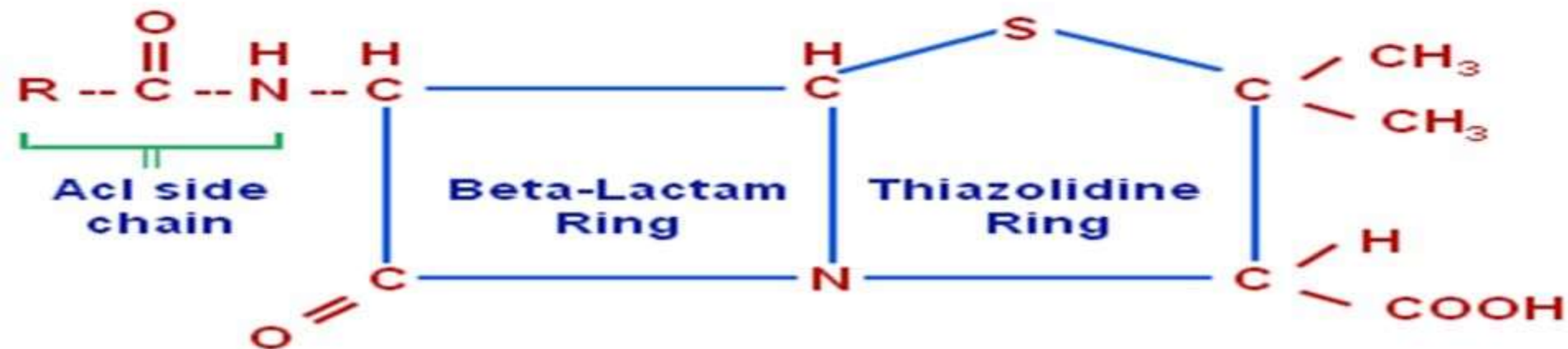
Penicillins as well as cephalosporins are called **beta-lactam antibiotics** and are characterized by **three** fundamental structural requirements:

- **The fused beta-lactam & Thiazolidine ring structure.**
- **A free carboxyl acid group.**
- **One or more substituted amino acid side chains.**



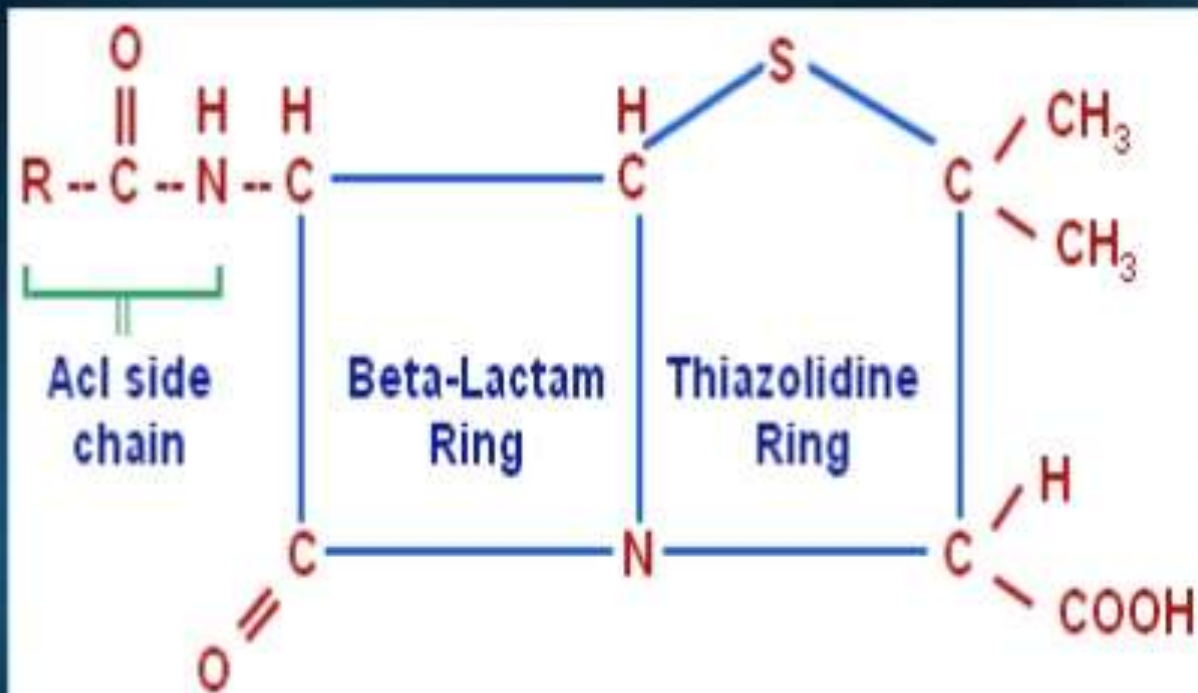
# Chemistry

- Penicillin nucleus consists of
  - Thiazolidine ring (Ring A)-
    - Sulphur containing with COOH (Carboxyl group),
  - Beta lactam ring (Ring B) – (Broken by Betalactamase)
    - Side chain is attached at position – 6- (NHCOR)
- Side chains attached through amide linkage. (Broken by Amidase)



# Synthetic antibiotics:

*general structure of penicillin*

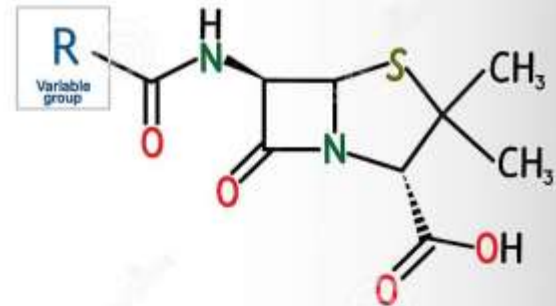


**General Structure of Penicillins**

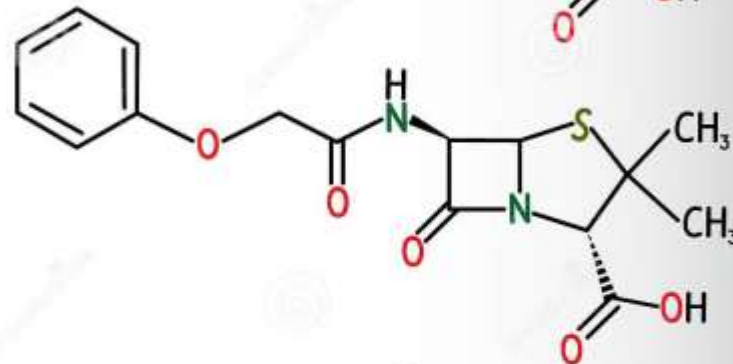


# Natural penicillins

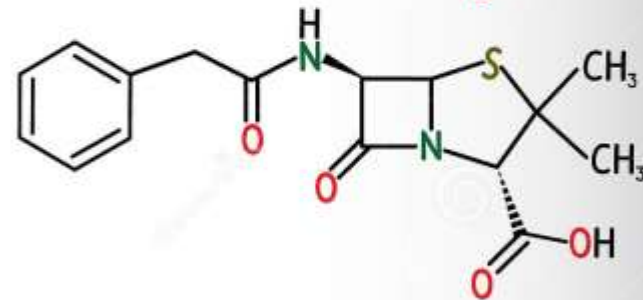
Penicillin



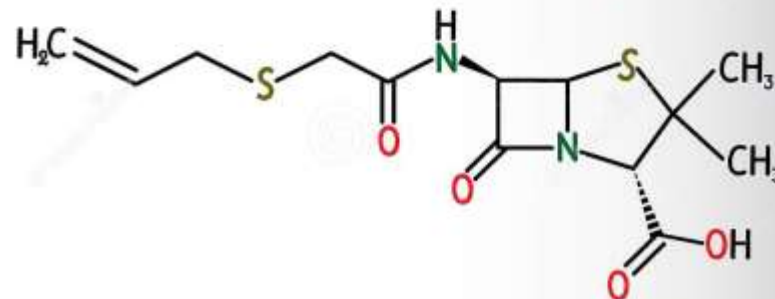
Penicillin V



Penicillin G



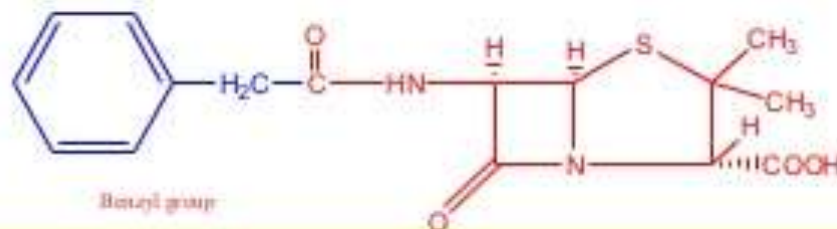
Penicillin O



## Penicillin G (Benzylpenicillin)

- Penicillin G is also referred to as **gold standard penicillin**.
- Penicillin G is not acid resistant it is acid sensitive.
- 3 reasons for the acid sensitivity of penicillin G.
  1. **Ring strain**. ( 4 membered betalactam ring + 5 membered thiazolidine ring) As a result penicillins suffers large angle and torsional strains. Acid catalyzed ring opening relieves these strains by breaking open the more highly  $\beta$ -lactam ring.
  2. **Highly reactive carbonyl group**. The resonance stabilization is impossible for the  $\beta$ -lactam ring because of the increase in angle strain that would result in having a double bond within  $\beta$ -lactam ring. So the angle of the  $\beta$ -lactam ring constrained to  $90^\circ$ . So the lone pair is localized on the N atom, and the carbonyl group is more electrophilic than one would expect for a tertiary amide.
- **Influence of the acyl side chain**: Acyl group open up the lactam ring . So Penicillin G has a self-destruct mechanism built in its structure.

Benzylpenicillin is broken down by stomach acid and destroyed by staphylococcus penicillinase. So it can be given by IV.



### Therapeutic uses:

Drugs of the penicillin group are effective for infections caused by Gram-positive bacteria (streptococcus, pneumococcus, and others), spirochaetae, and other pathogenic microorganisms.

**Drugs of this group are ineffective with respect to viruses, mycobacteria tuberculosis, fungi, and the majority of Gram-negative microorganisms.**

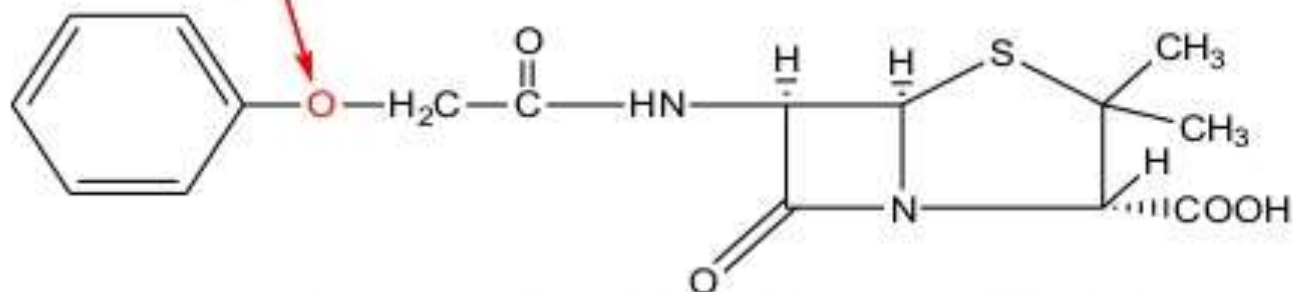
Benzylpenicillin is the drug of choice for infections caused by sensitive organisms. This includes streptococci infections (except enterococci), gonococci, and meningococci that do not produce beta-lactam anaerobes.

Benzylpenicillin is used for croupous and focal pneumonia, skin infections, soft tissue and mucous membranes, peritonitis, cystitis, syphilis, diphtheria, and other infectious diseases.



## PENICILLIN- V

It withdraws the electrons away from the carbonyl oxygen and reduce the tendency to act as a nucleophile



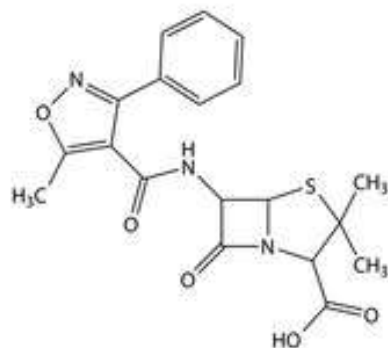
Phenoxy methyl Penicillin ( Penicillin V)

- By placing electron withdrawing group in the side chain which could draw electrons away from the carbonyl oxygen and reduce its tendency to act as a nucleophile.
- Penicillin- V has electron withdrawing oxygen on the acyl side chain with electron withdrawing effect. It has more acid stability than penicillin G .
- It is more stable in acid in the stomach, so it can be given orally.
- In fact acid sensitivity can be solved by having an electron withdrawing group on the Acyl side chain.

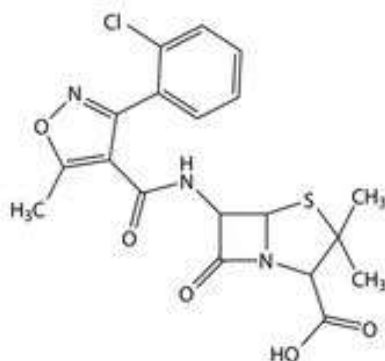




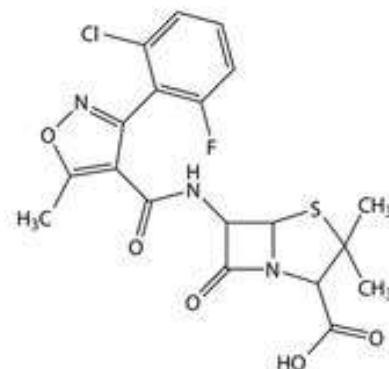
# Isoxazole penicillins. Part 6



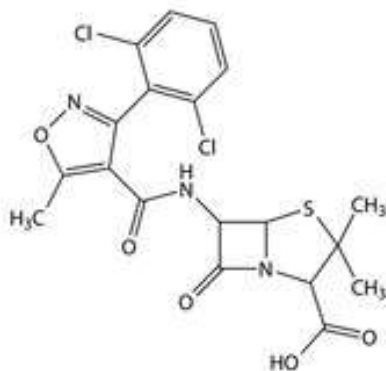
Oxacillin



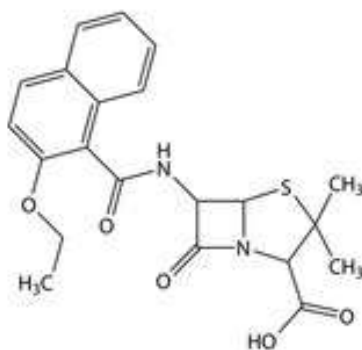
Cloxacillin



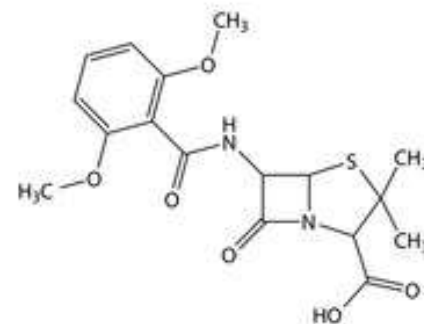
Flucloxacillin



Dicloxacillin

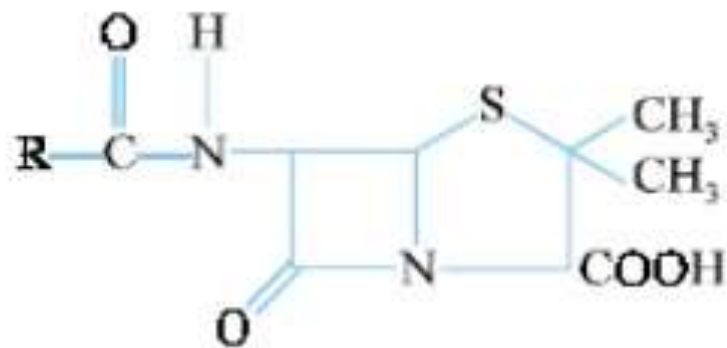


Nafcillin



Meticillin

# Penicillins



Name	R	Name	R
Ampicillin		Cloxacillin Sodium	
Amoxicillin Trihydrate		Methicillin Sodium	
Ciclacillin			

## Structural Activity Relationship (SAR)

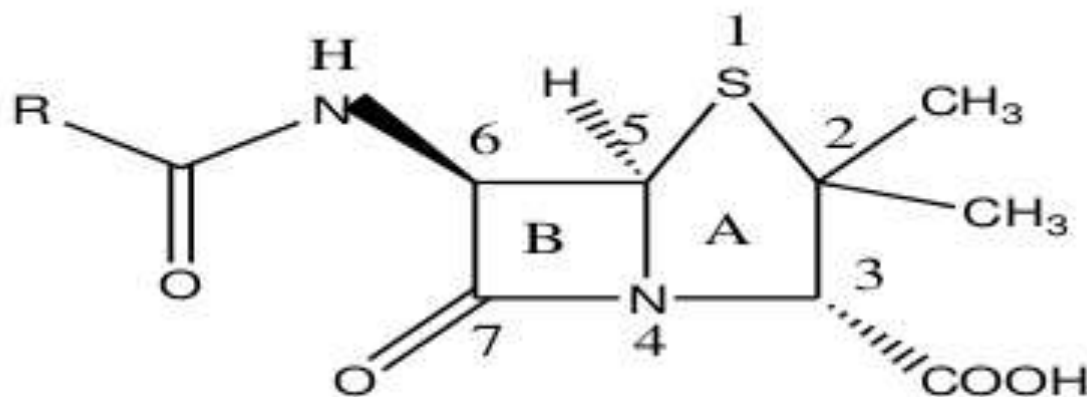
**Position 1** – When the sulfur atom of the Thiazolidine ring is oxidized to a sulfone or sulfoxide, it improves acid stability, but decreases the activity of the agent.

**Position 2** – No substitutions allow at this position, any change will lower activity. The methyl groups are necessary

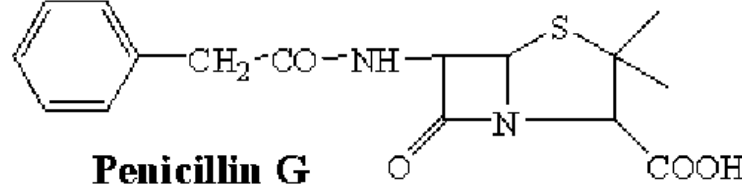
**Position 3** – The carboxylic acid of the Thiazolidine is required for activity. If it is changed to an alcohol or ester, activity is decreased.

**Position 4** – The nitrogen is a must.

**Position 5** – No substitutions allowed.

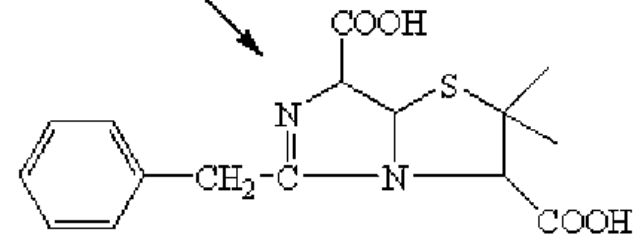
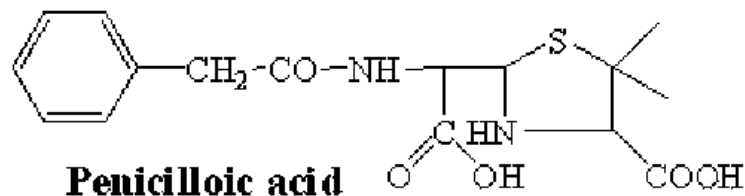






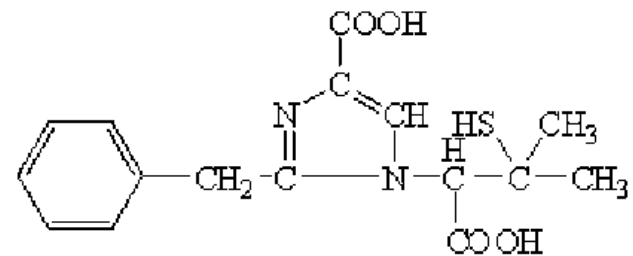
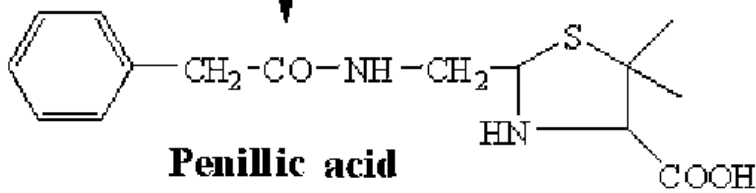
weak acid or  $\beta$ -lactamase  
75 °C

weak acid 75 °C



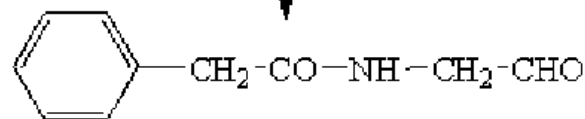
$H^+$  75 °C

$OH^-$  or anaerobic condition  
ambient temperature



$H^+$  75 °C

$H^+$  ambient temperature



aerobic condition  
ambient temperature

unknown products

# DIFFERENT CLASSES OF ANTIBIOTICS - AN OVERVIEW

**Key:** ● COMMONLY ACT AS BACTERIOSTATIC AGENTS, RESTRICTING GROWTH & REPRODUCTION ● COMMONLY ACT AS BACTERICIDAL AGENTS, CAUSING BACTERIAL CELL DEATH

