

SNS COLLEGE OF PHARMACY AND HEALTH SCIENCES

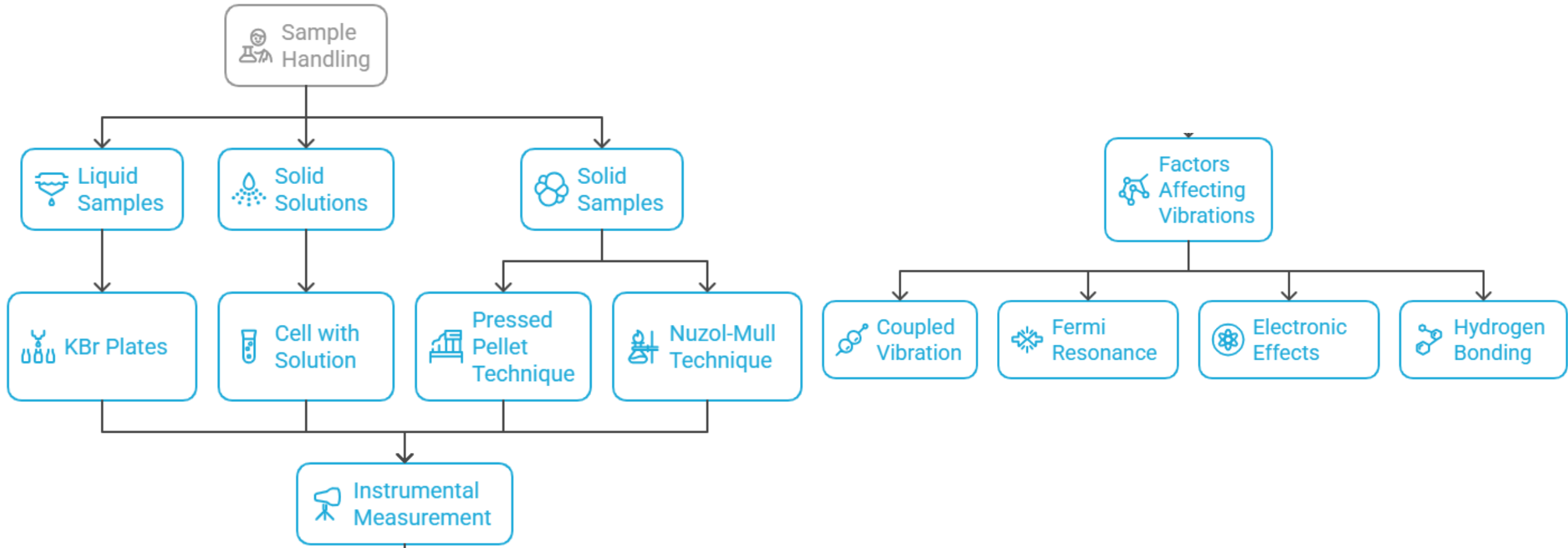
Affiliated To The Tamil Nadu Dr. MGR Medical University, Chennai
Approved by Pharmacy Council of India, New Delhi.
Coimbatore -641035

COURSE NAME: INSTRUMENTAL METHODS OF ANALYSIS (BP 701 T)

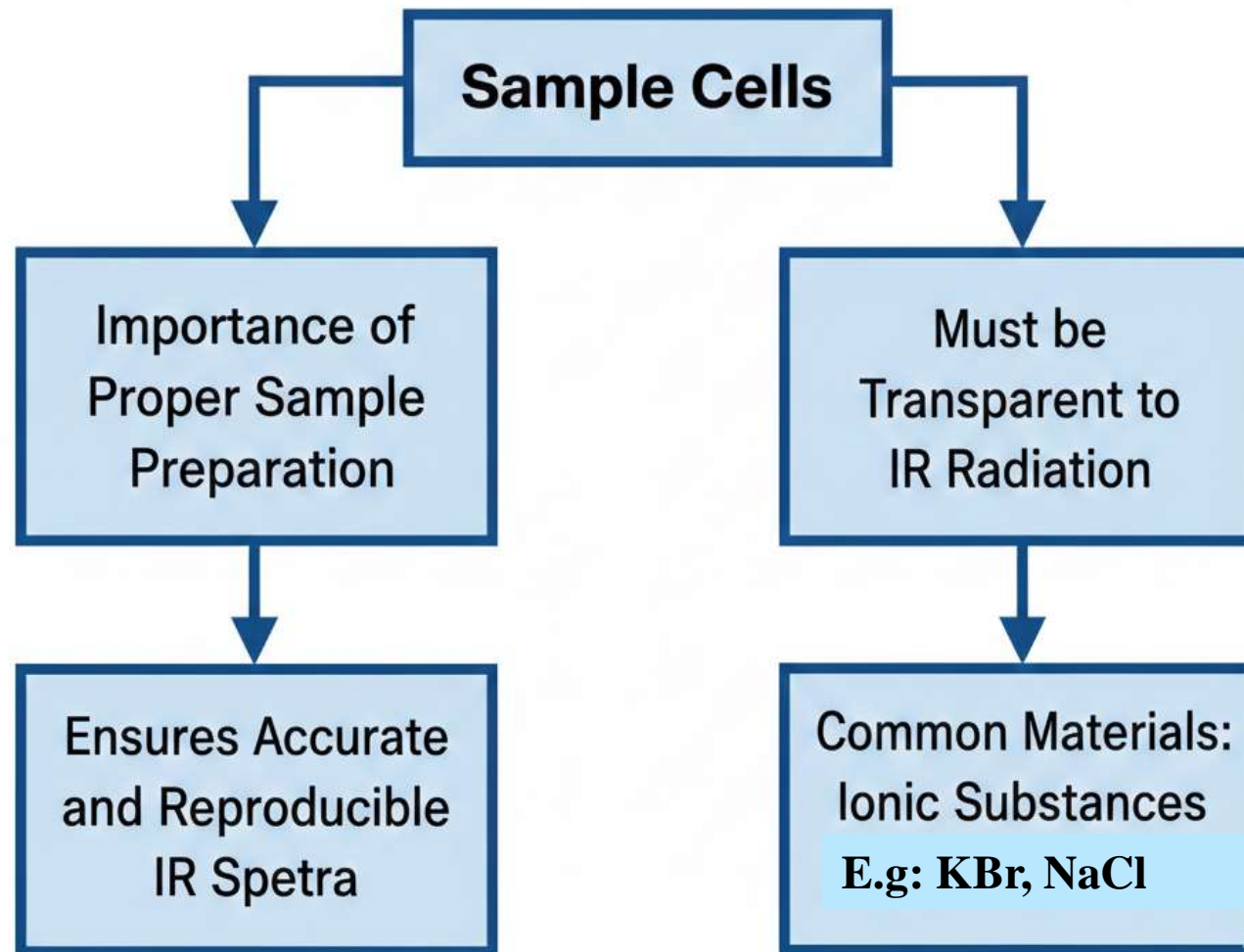
VII SEM/ IV YEAR

TOPIC 12: SAMPLE HANDLING AND FACTORS AFFECTING VIBRATIONS

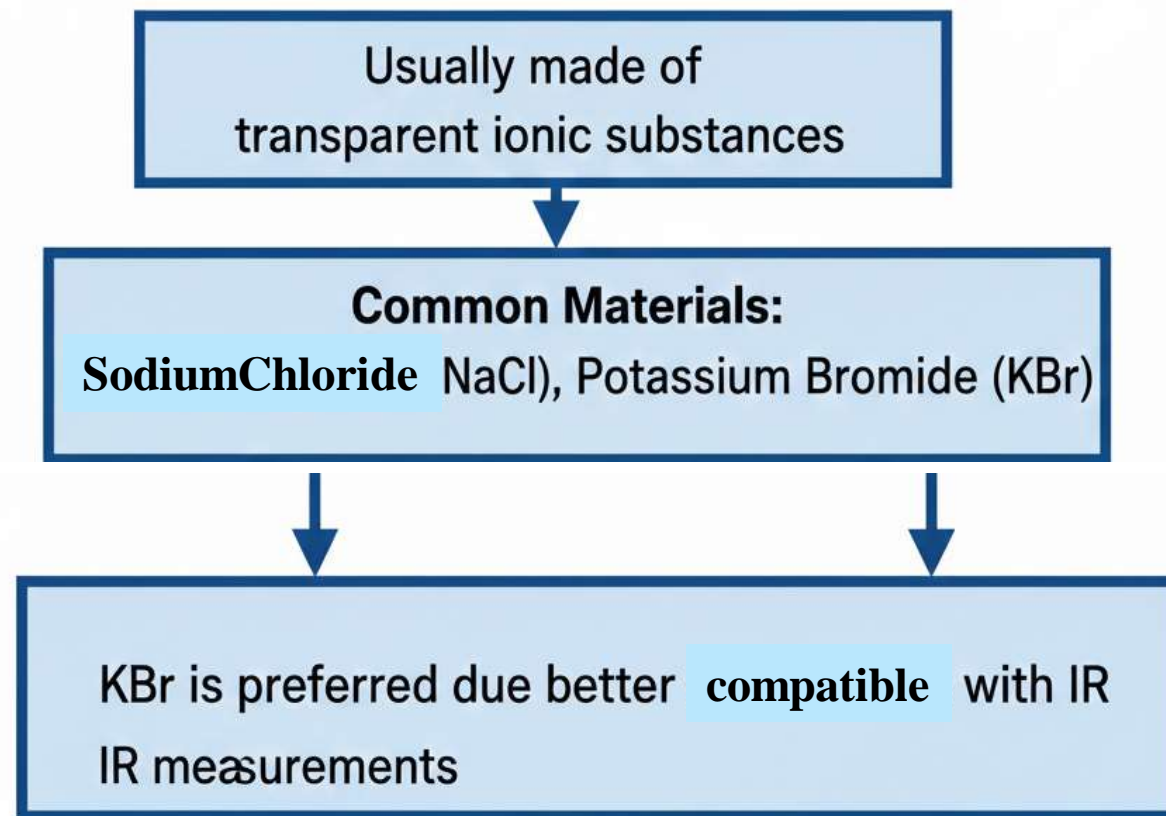
MINDMAP:



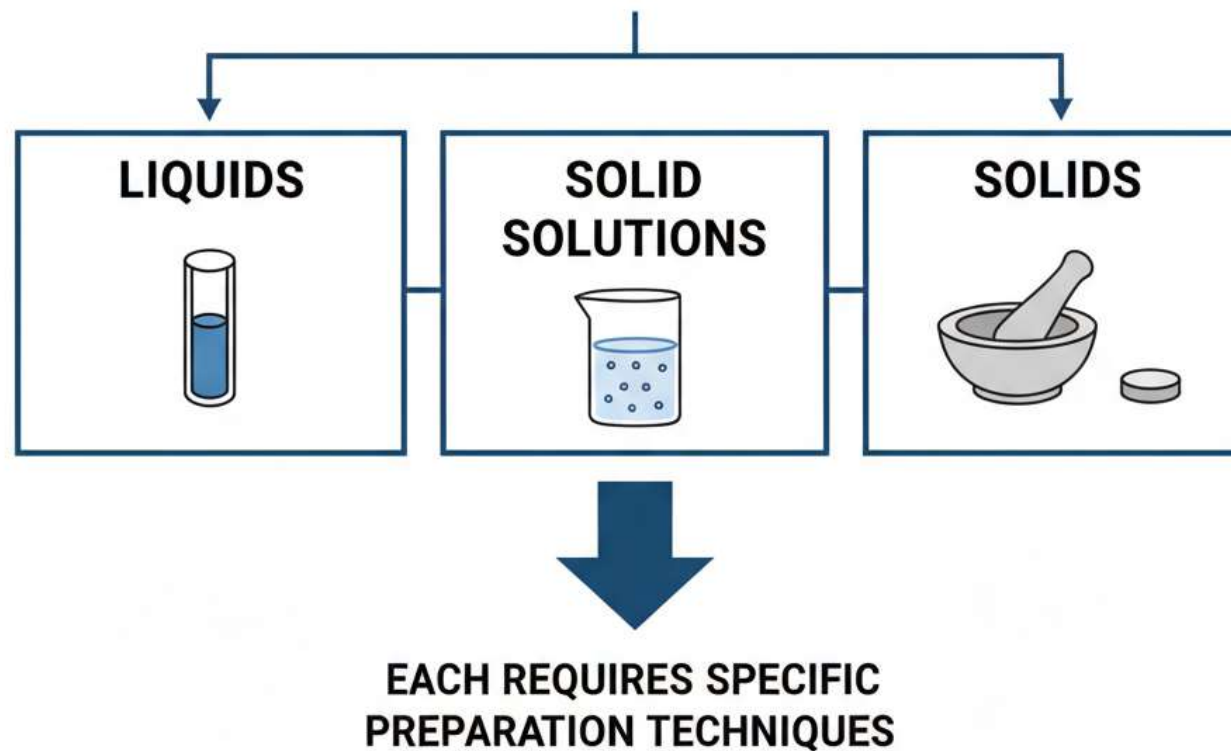
Introduction to Sample Handling in IR Spectroscopy



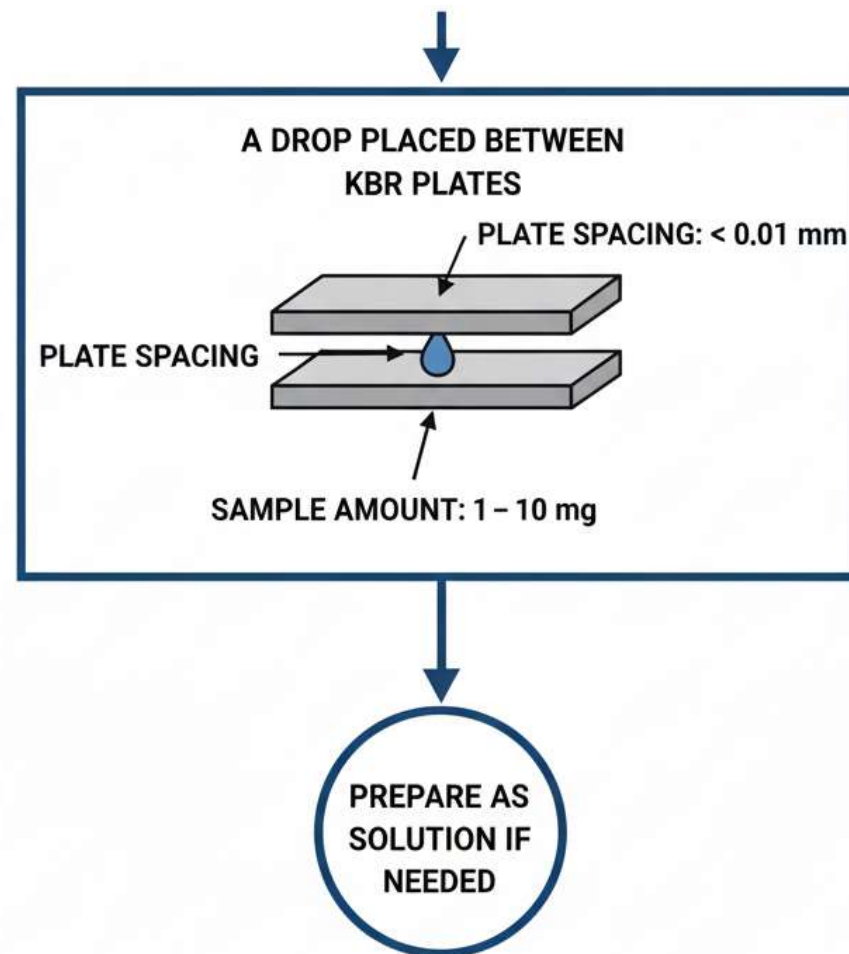
Materials for Sample Cells



Forms of Infrared Samples



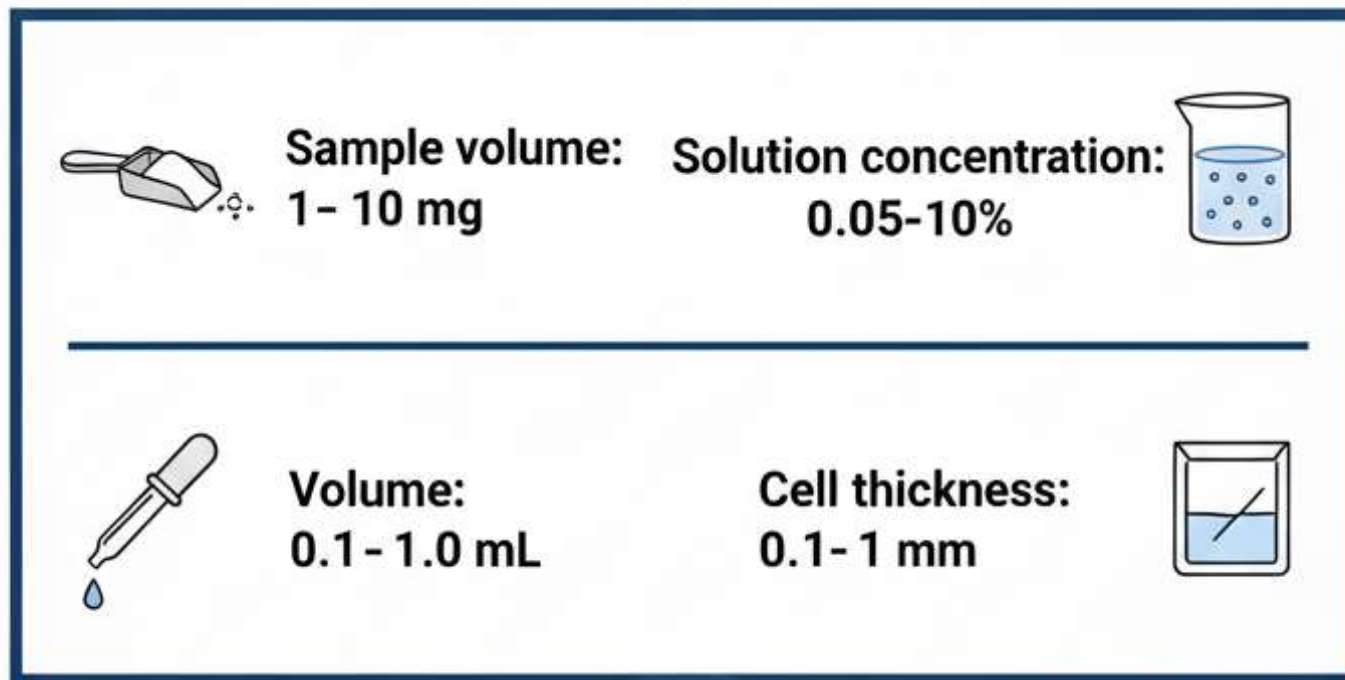
Handling Liquid Samples



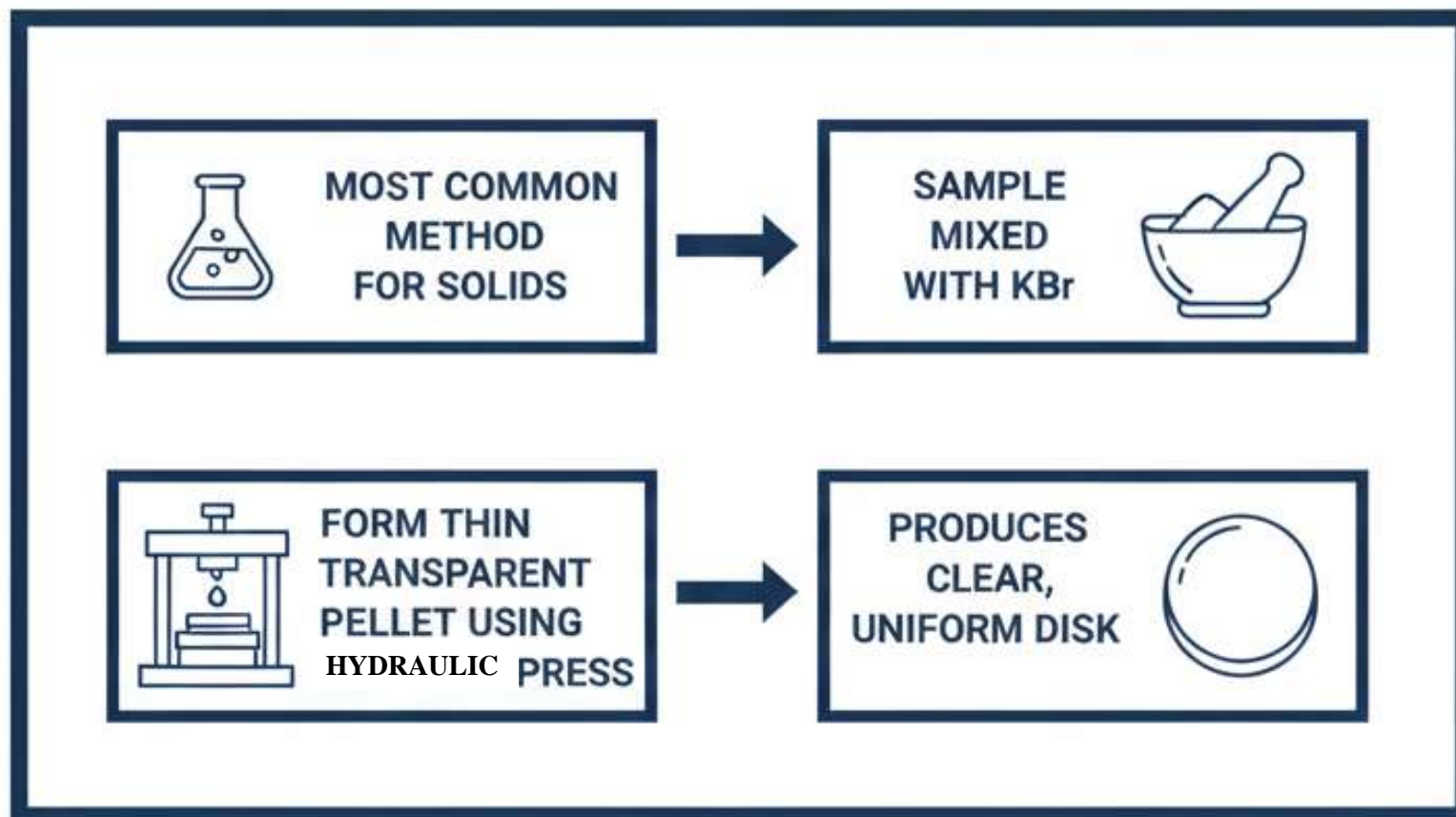
Considerations for Liquid Samples



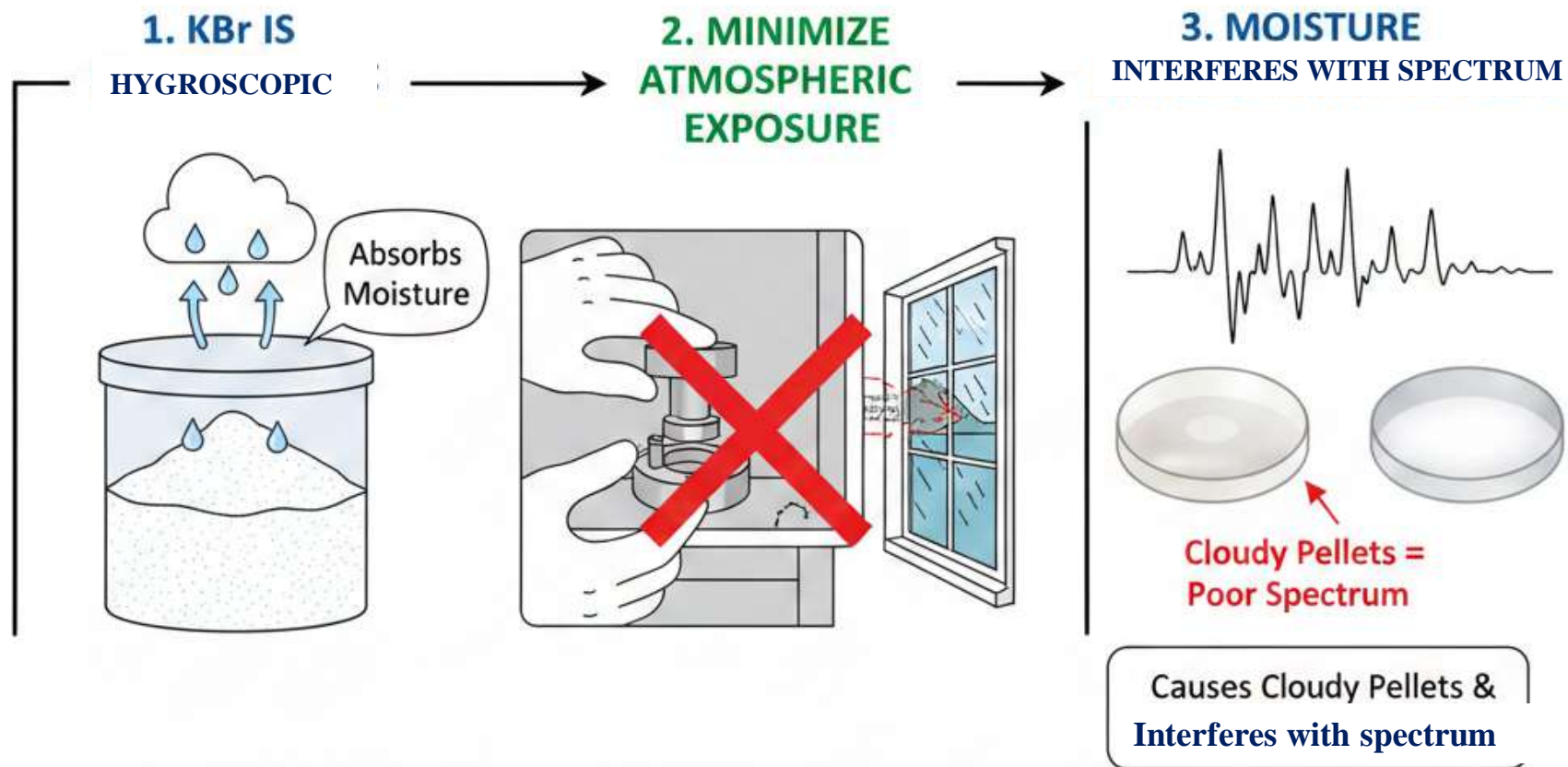
Solid Solutions



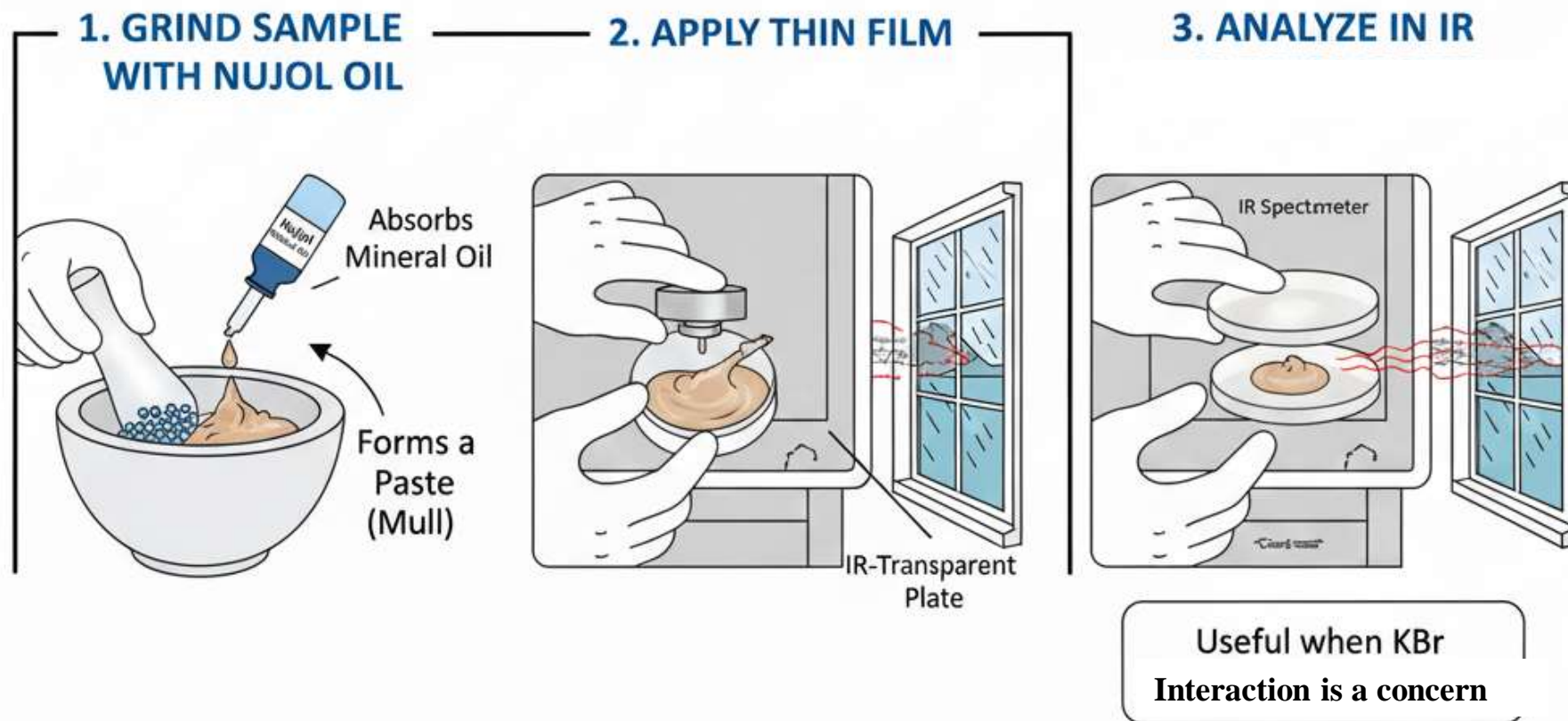
Handling Solid Samples – Pressed Pellet Technique



Precautions with KBr Pellets



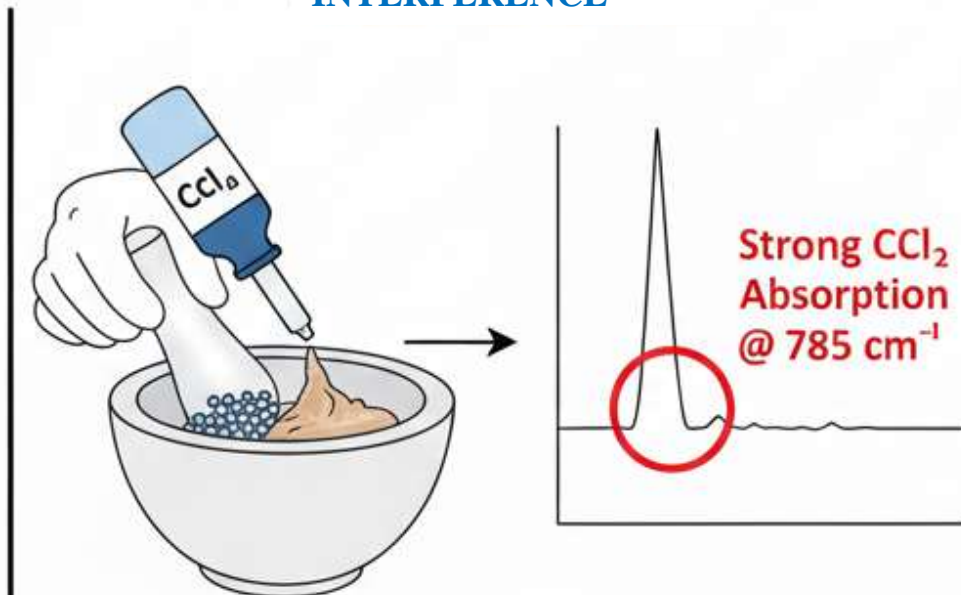
Nujol Mull (Mull Technique)



Caution with Solvents in Mull Technique

1. CCl_2 ABSORPTION

INTERFERENCE



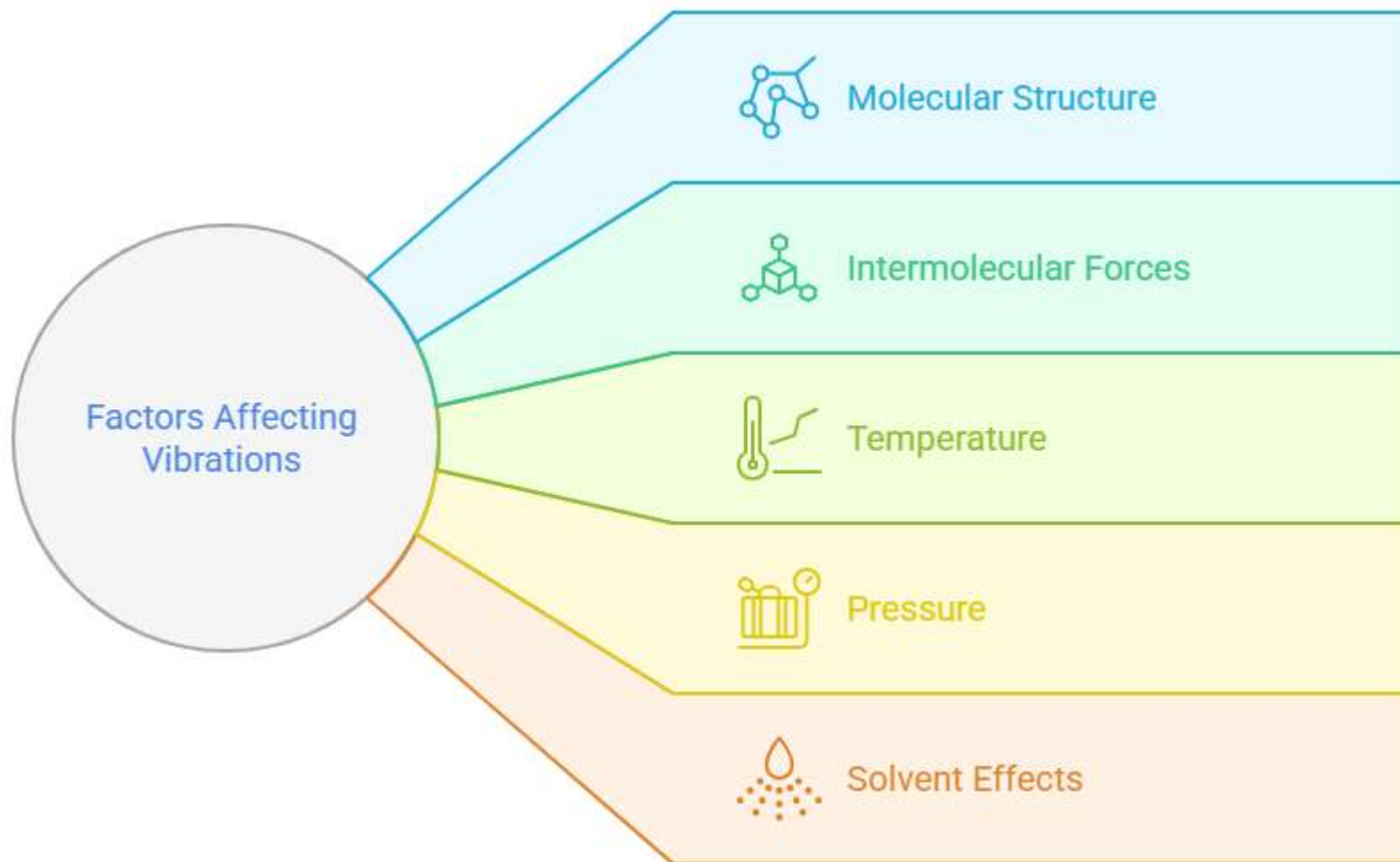
CCl_2 absorbs strongly, Interfering with spectrum analysis.

2. AVOID INSTANT REACTIONS



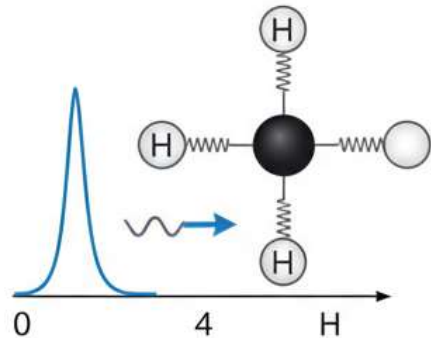
Avoid solvent-sample combinations that react instantly

Factors Affecting Vibrations



Coupled Vibrations

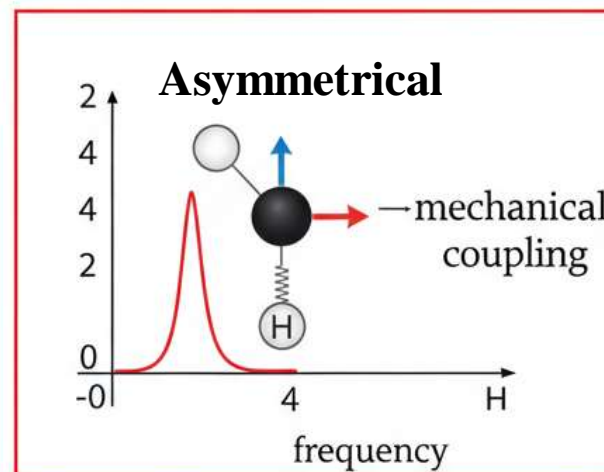
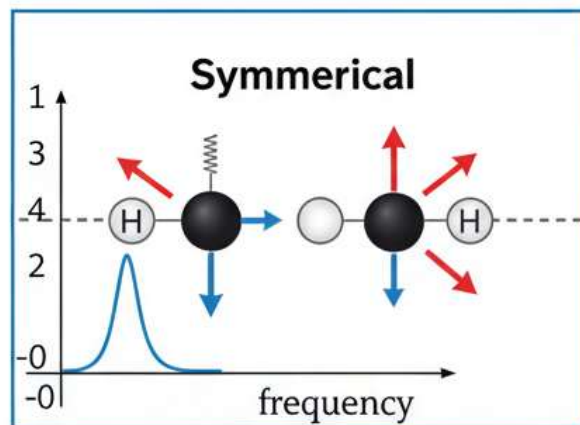
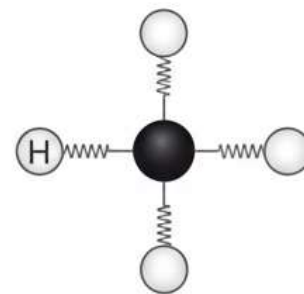
Isolated C-H:



Single stretching frequency

Methylene group

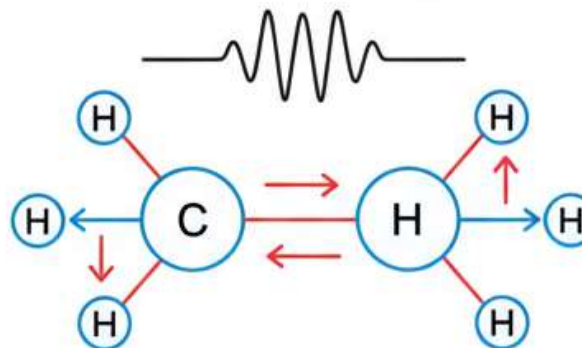
(CH₂-):



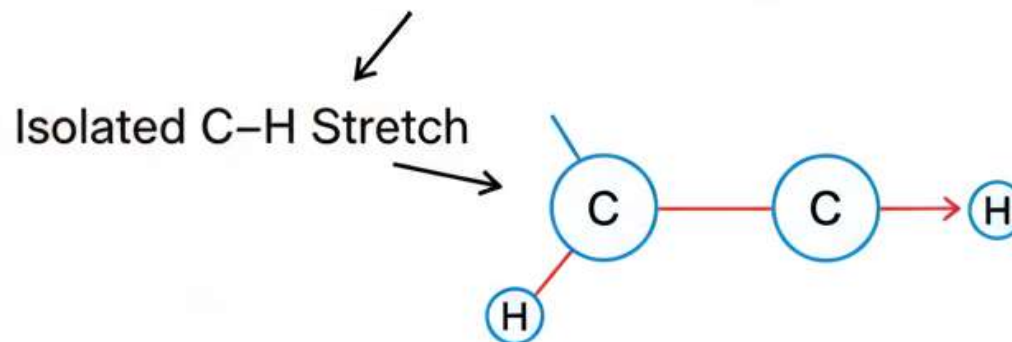
Two stretching modes
Due to mechanical coupling

Coupled Vibrations

Asymmetrical Stretching: Higher Frequency



Symmetrical Stretching: Lower Frequency



Occur at different frequencies
than isolated C-H

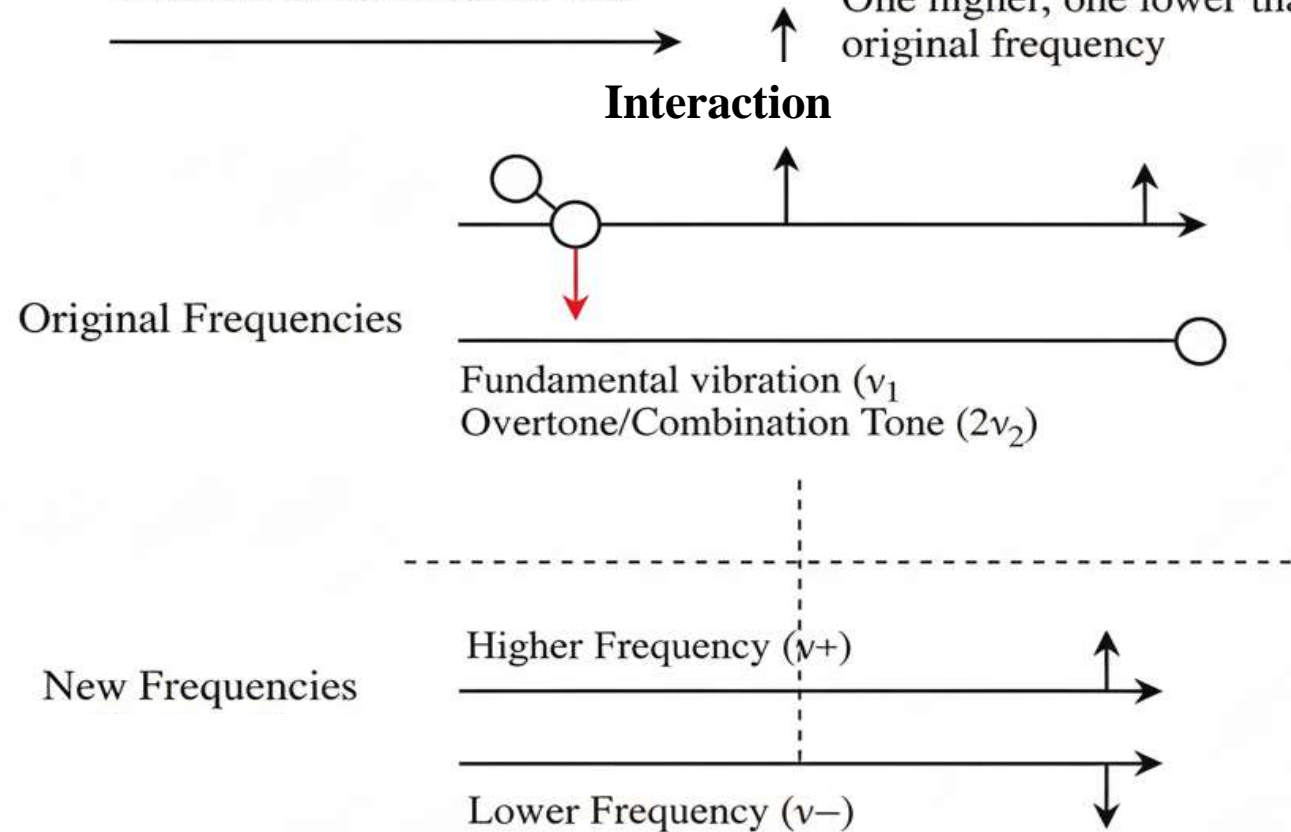
Fermi Resonance

Interaction between:

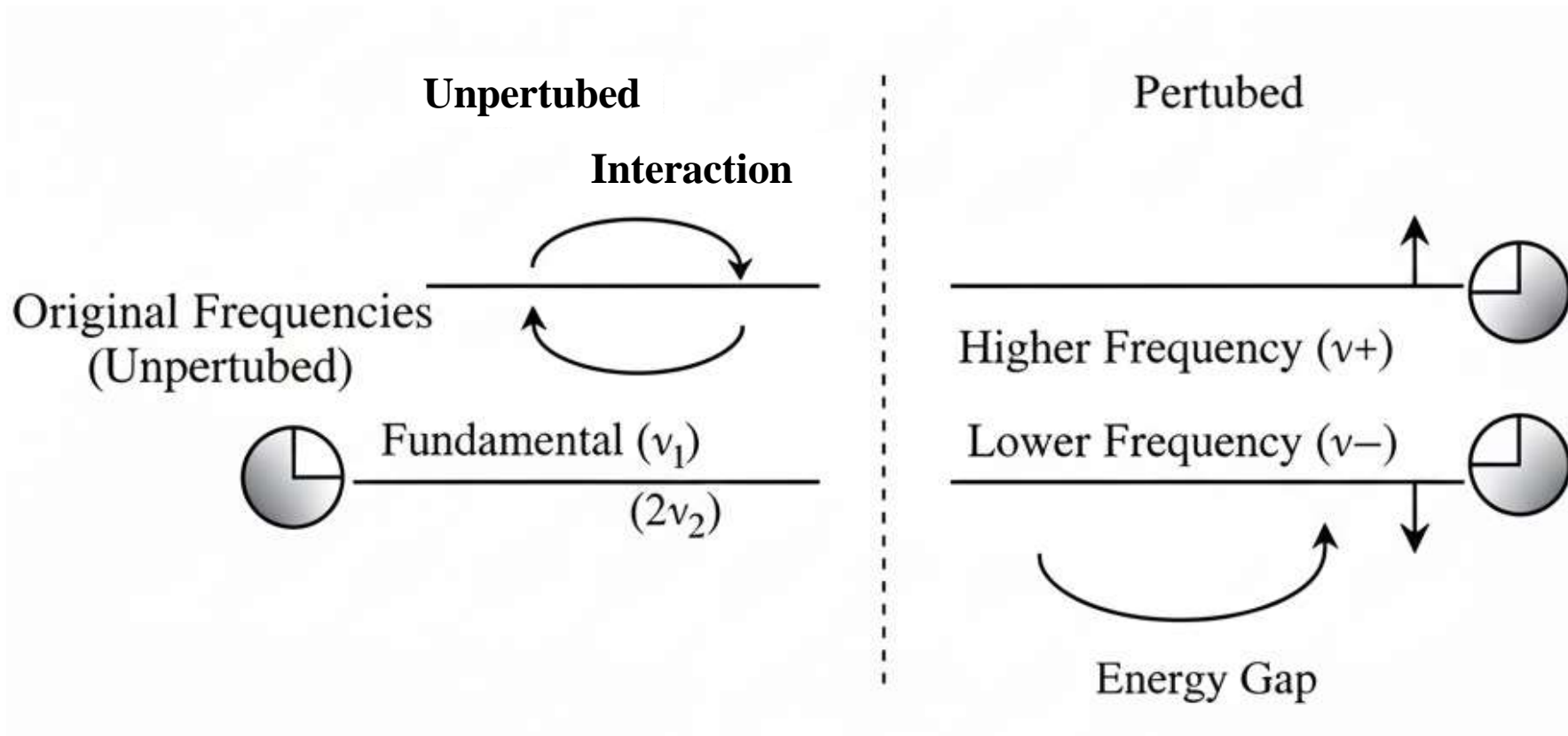
Fundamental vibration
Overtone or combination tone

Produces two new vibrational modes

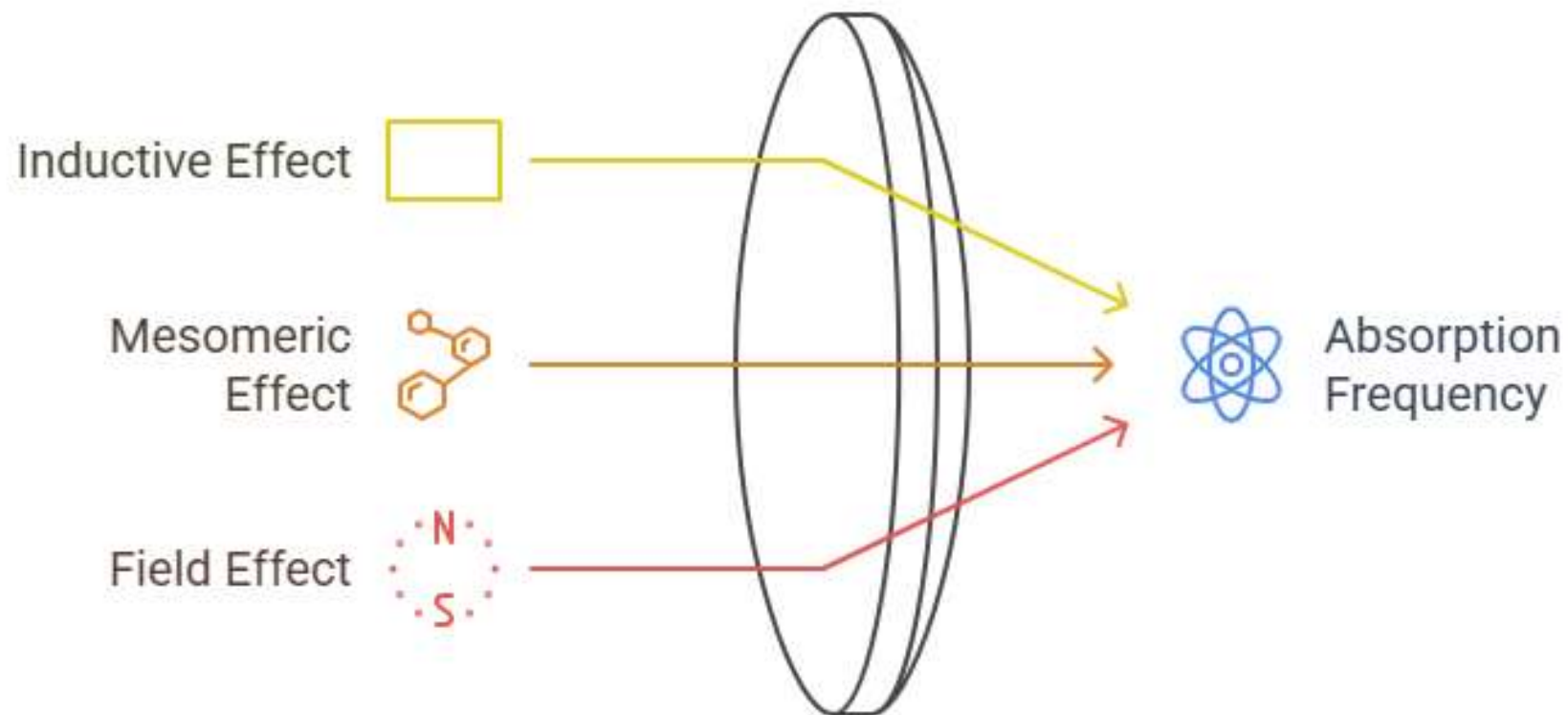
One higher, one lower than
original frequency



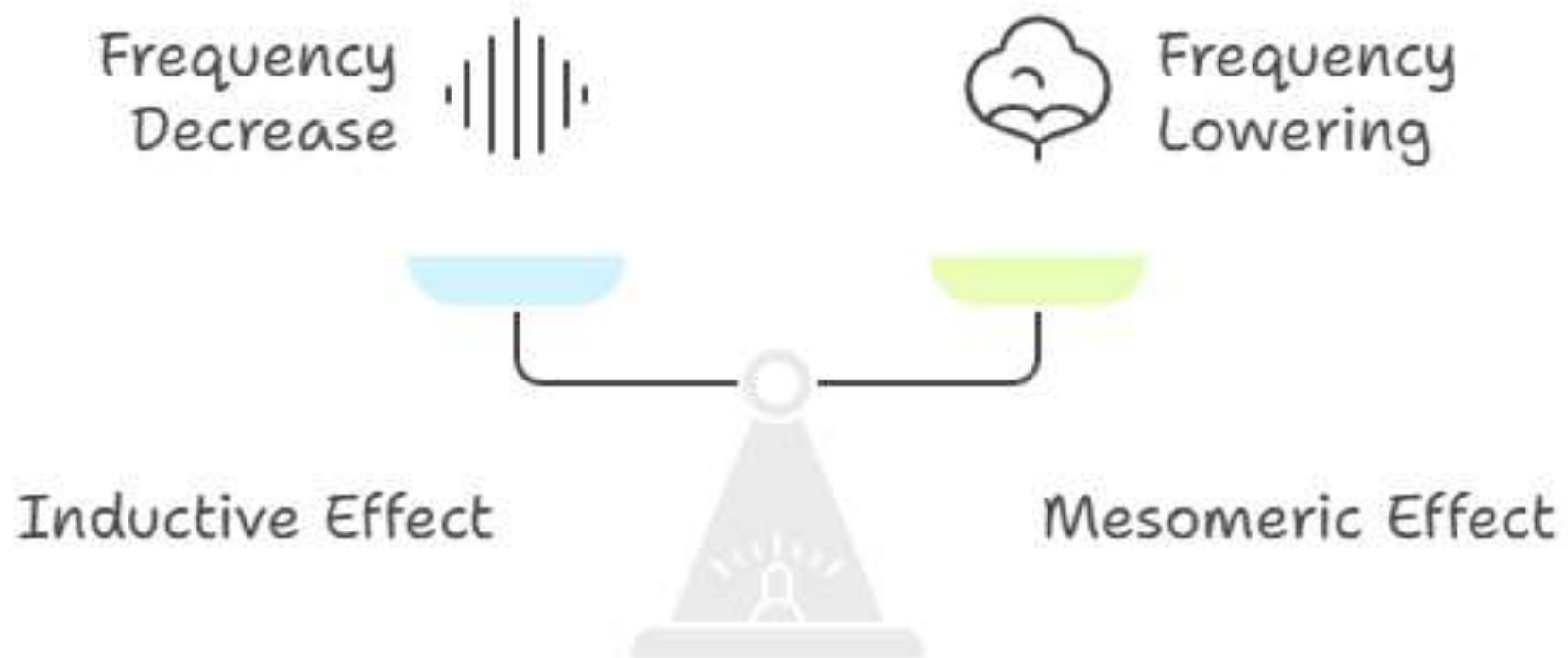
Fermi Resonance



Electronic Effects



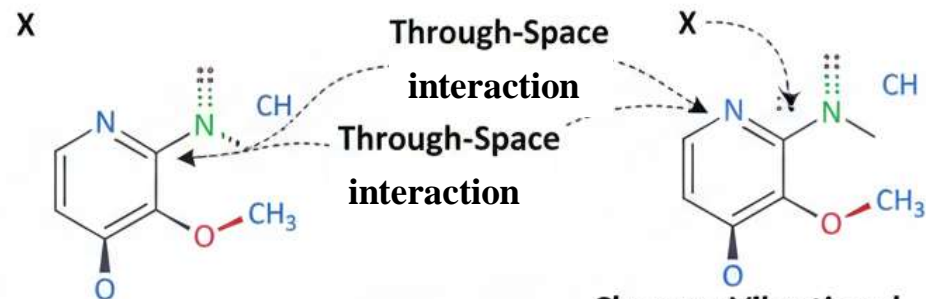
Inductive & Mesomeric effect



Field effect

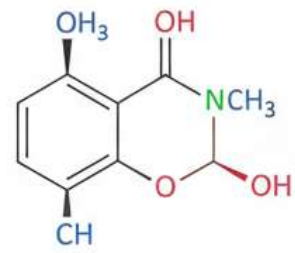
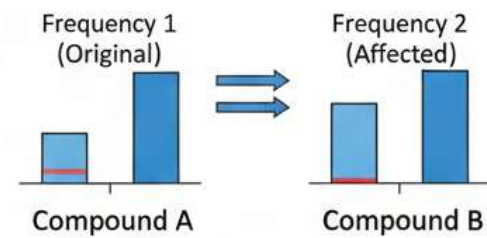
Ortho-Haloacetophene A

Ortho-Haloacetophene B



Changes Vibrational Frequencies

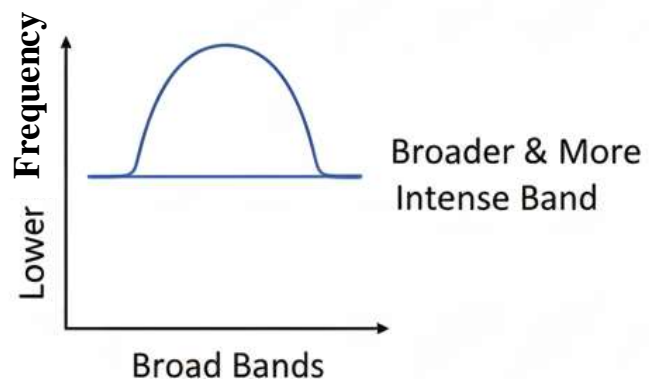
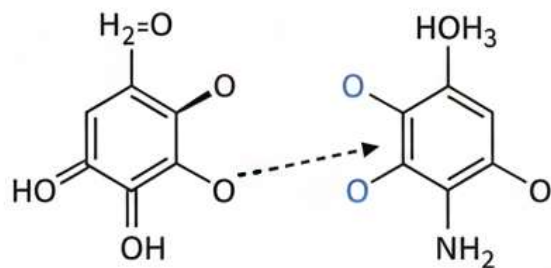
Lone Pairs Influence Other Through Space



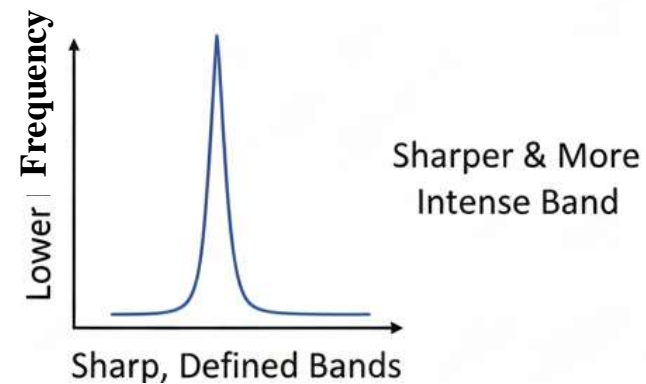
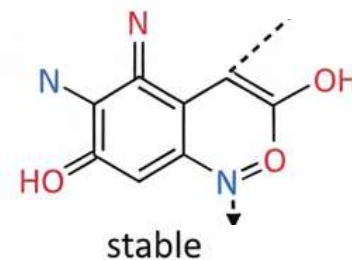
Hydrogen Bonding

Occurs Between Proton Donor and Acceptor
 Stronger H-bond → Longer O–H bond

Intermolecular



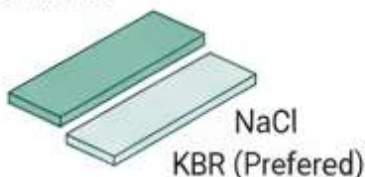
Intramolecular Hydrogen Bonding



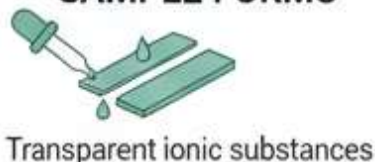
Summary:

SAMPLE HANDLING

CELLS



SAMPLE FORMS



Liquids:

- KBR plates (<0.01mm)
- Anhydrous Organic Solvents (e.g., CH₂Cl₂)
- Avoid H₂O



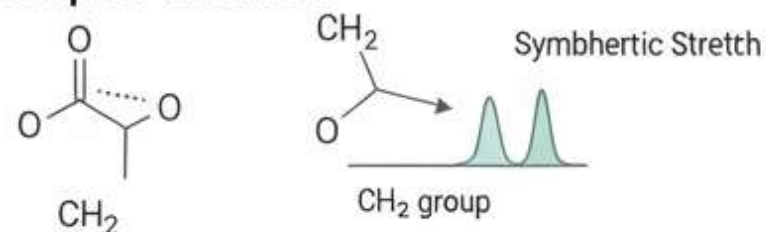
Solid Solutions:

- 1-10mg solid + CCl₄ (0.1-1ml)
- 0.05-10% solution

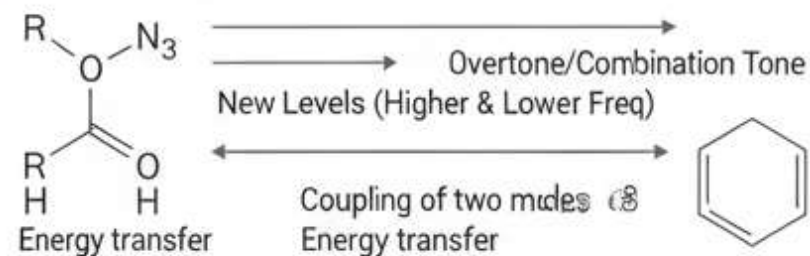


FACTORS AFFECTING VIBRATIONS

Coupled Vibration



Fermi Reseance



Summary:

SAMPLE HANDLING

• Solids



Pressed Pellet (KBR + Sample)
Thin transparent
Thin film on cell

—————→ KBR is hygroscopic

• Solids



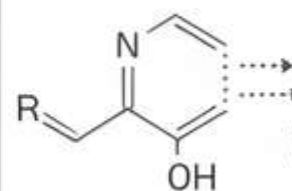
Mull Technique
Thin film on cell)



Mull Technique
(Nujol + Sample)

FACTORS AFFECTING VIBRATIONS

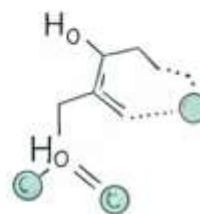
• Electronic Effects



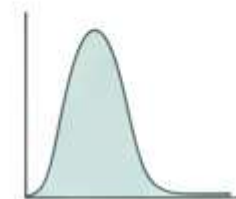
Mesomeric effect = Lower Higher Frq
(Bond Weakening)

Field Effects Effect = Lower Frq

• Hydrogen Bonding



Intermolecular
(Broad Band)
(Sharp Band)



Salicylic acid
Proton Donor + Acceptor
Stronger H-bond = Lower Frq,
Broader, More Intense

Assessment

1. Which solvent should be used for analyzing liquid samples?



Assessment

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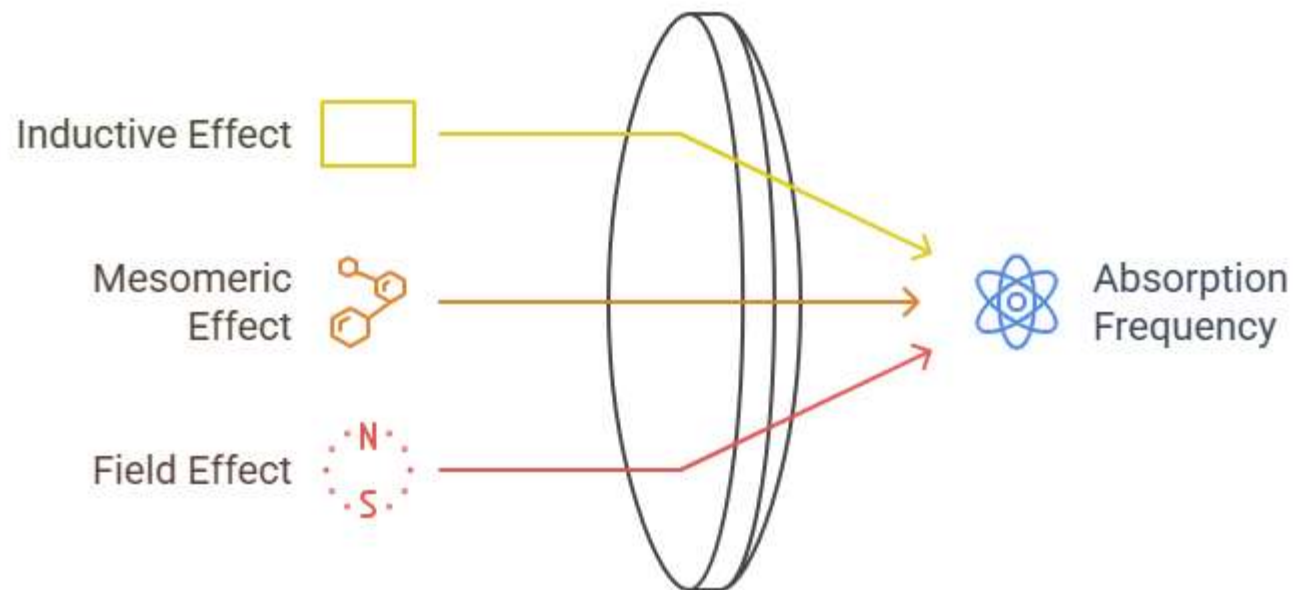
Assessment

2. In which type of electronic effect the lone pairs influence each other through space?



Assessment

2. In which type of electronic effect the lone pairs influence each other through space?



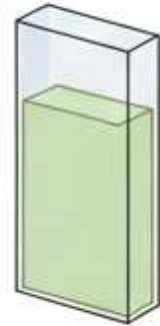
Assessment

3. Which material is preferred for Sample Cells?

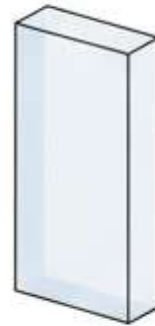


Assessment

3. Which material is preferred for Sample Cells?



NaCl



Quartz



Plastic

References

1. Pavia DL, Lampman GM, Kriz GS, Vyvyan JR. Introduction to spectroscopy. 5th ed. Boston (MA): Cengage Learning; 2015.
2. Silverstein RM, Webster FX, Kiemle DJ, Bryce DL. Spectrometric identification of organic compounds. 8th ed. Hoboken (NJ): John Wiley & Sons; 2015.
3. Nakanishi K, Solomon PH. Infrared absorption spectroscopy. 2nd ed. San Francisco (CA): Holden-Day; 1977.
4. Colthup NB, Daly LH, Wiberley SE. Introduction to infrared and Raman spectroscopy. 3rd ed. San Diego (CA): Academic Press; 1990.
5. Banwell CN, McCash EM. Fundamentals of molecular spectroscopy. 4th ed. London: McGraw-Hill; 1994.

Thank
you!