

SNS COLLEGE OF PHARMACY AND HEALTH SCIENCES

Affiliated To The Tamil Nadu Dr. MGR Medical University, Chennai

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Coimbatore -641035

COURSE NAME : PHARMACEUTICAL MICROBIOLOGY - BP303 T

B.PHARM II YEAR / III SEM

UNIT 3

SUB TOPIC :DISINFECTANTS - CLASSIFICATION AND MODE OF ACTION

Chemical Type (Chemical Families)

Halogens	Chlorine (bleach, hypochlorites) Iodine (iodophors)
Alcohols	Ethanol, Isopropanol (effective against many microbes, but spores)
Phenolics	Phenol, cresols (disrupt cell membranes)
Aldehydes	Formaldehyde, Glutaraldehyde (high-level disinfectants/sterilants)
Quaternary Ammonium Compounds	Benzalkonium chloride (common in household cleaners)
Oxidizing Agents	Hydrogen peroxide, peracetic acid (release reactive oxygen species)
Biguanides	Chlorhexidine (broad-spectrum, used as antiseptic)
Heavy Metals	Silver, copper (act by binding to proteins)

Antimicrobial Spectrum/Level

Low-Level

Kill most bacteria, some viruses, fungi
(e.g. QACs, diluted bleach)

Intermediate Level

Kill vegetative bacteria, mycobacteria, most viruses, but not spores
(e.g. alcohol, phenolic compounds)

High-Level

Kill all microorganisms except large numbers of bacterial spores
(e.g. glutaraldehyde, hydrogen peroxide)

Sporicidal

Kill all microorganisms, including spores
(e.g. glutaraldehyde, peracetic acid)

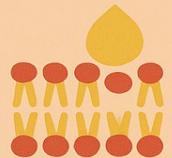
AIR DISINFECTANTS

Gaseous agents
like formaldehyde

DYES & DETERGENTS

Used as adjuncts
or mild disinfectants

COMMON MODES OF ACTION



Membrane damage

Disrupts the cell membrane's integrity, causing it to leak and leading to cell death.

Certain agents can increase membrane permeability, causing cytoplasmic contents to leak out.



Protein denaturation

Coagulates or denatures proteins, which are vital for cellular functions

Alcohols, phenols, and aldehydes work by denaturing proteins.



Oxidation

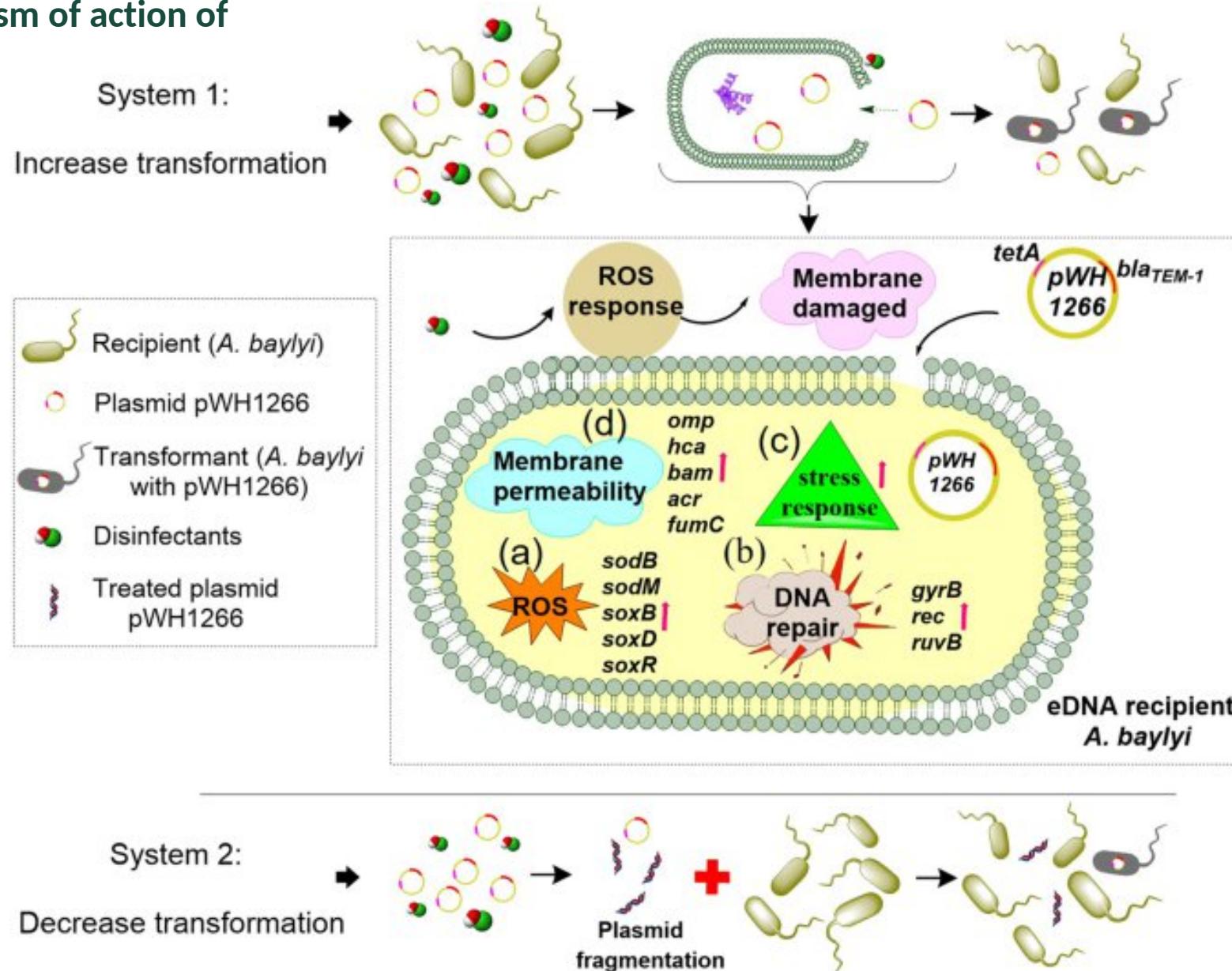
Strong oxidizing agents like chlorine and hydrogen peroxide can damage proteins, DNA, and other cellular components by oxidizing them.



Nucleic acid damage

Disinfectants can damage or inhibit the synthesis of DNA and RNA.

Mechanism of action of halogens



HOW HALOGENS WORK

CHLORINE

Forms hypochlorous acid (HOCl) in water, its active agent, which penetrates cells and oxidizes components.

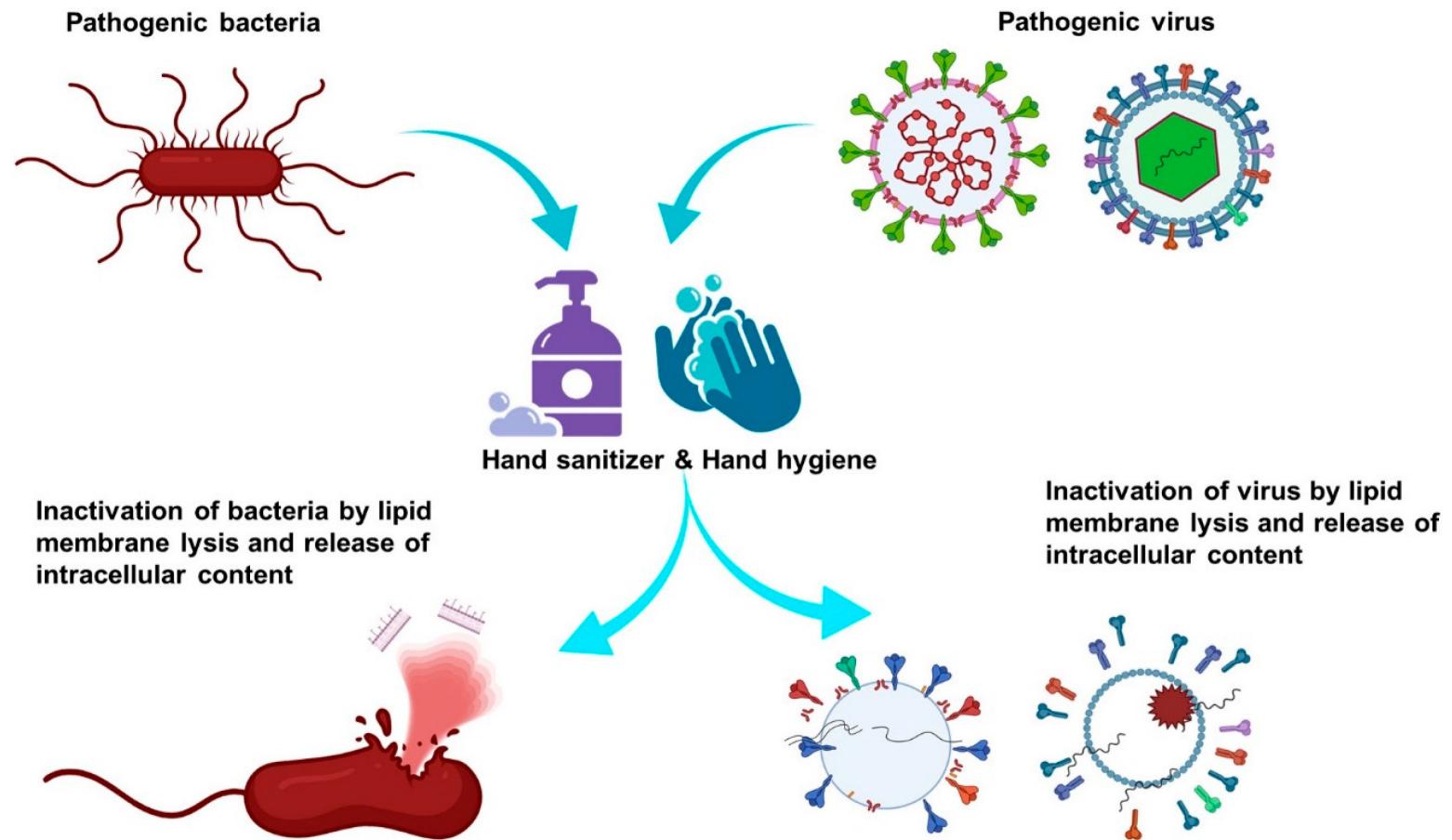
IODINE

Oxidizes proteins and lipids, destabilizing cell structures; used topically but can irritate skin

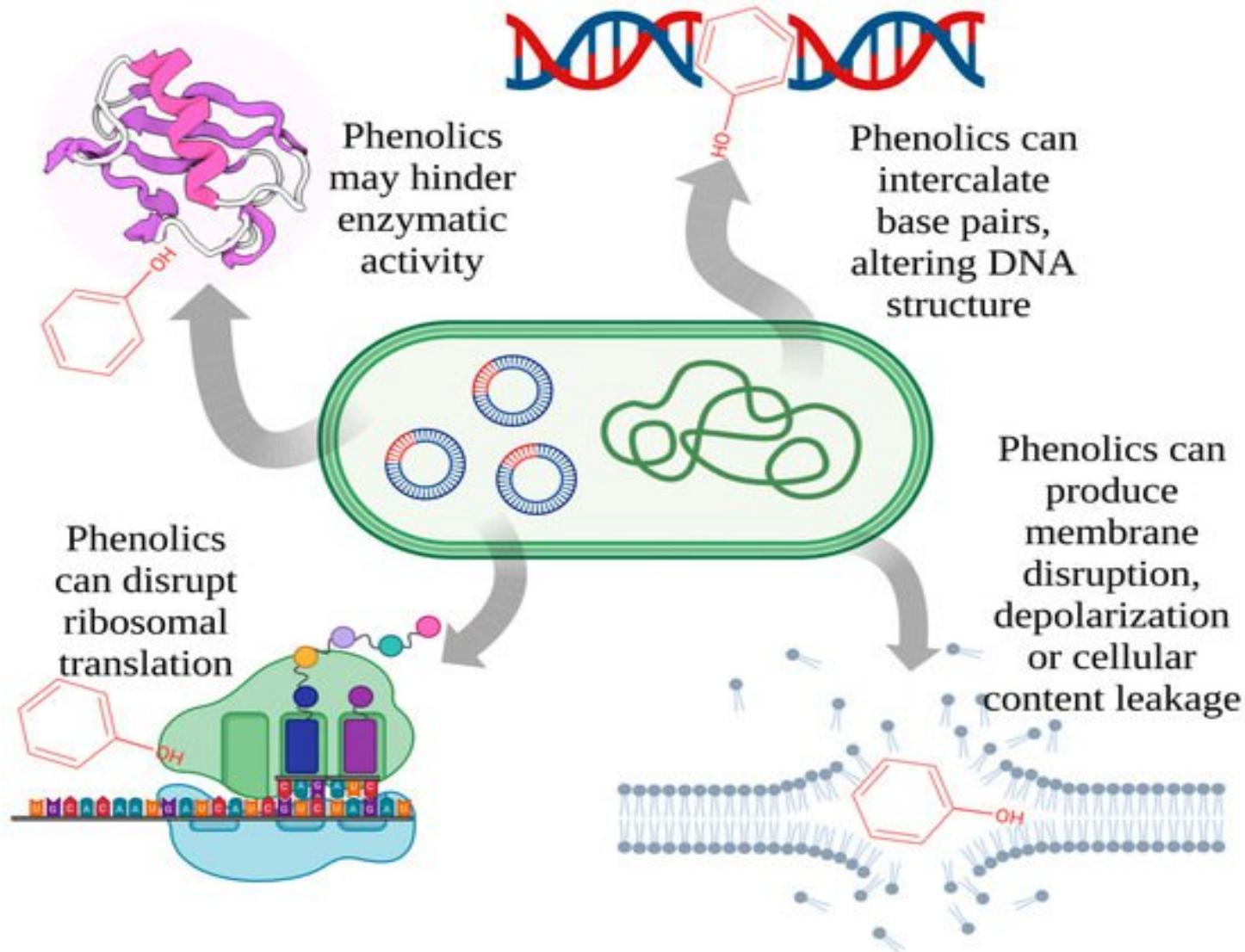
BROMINE

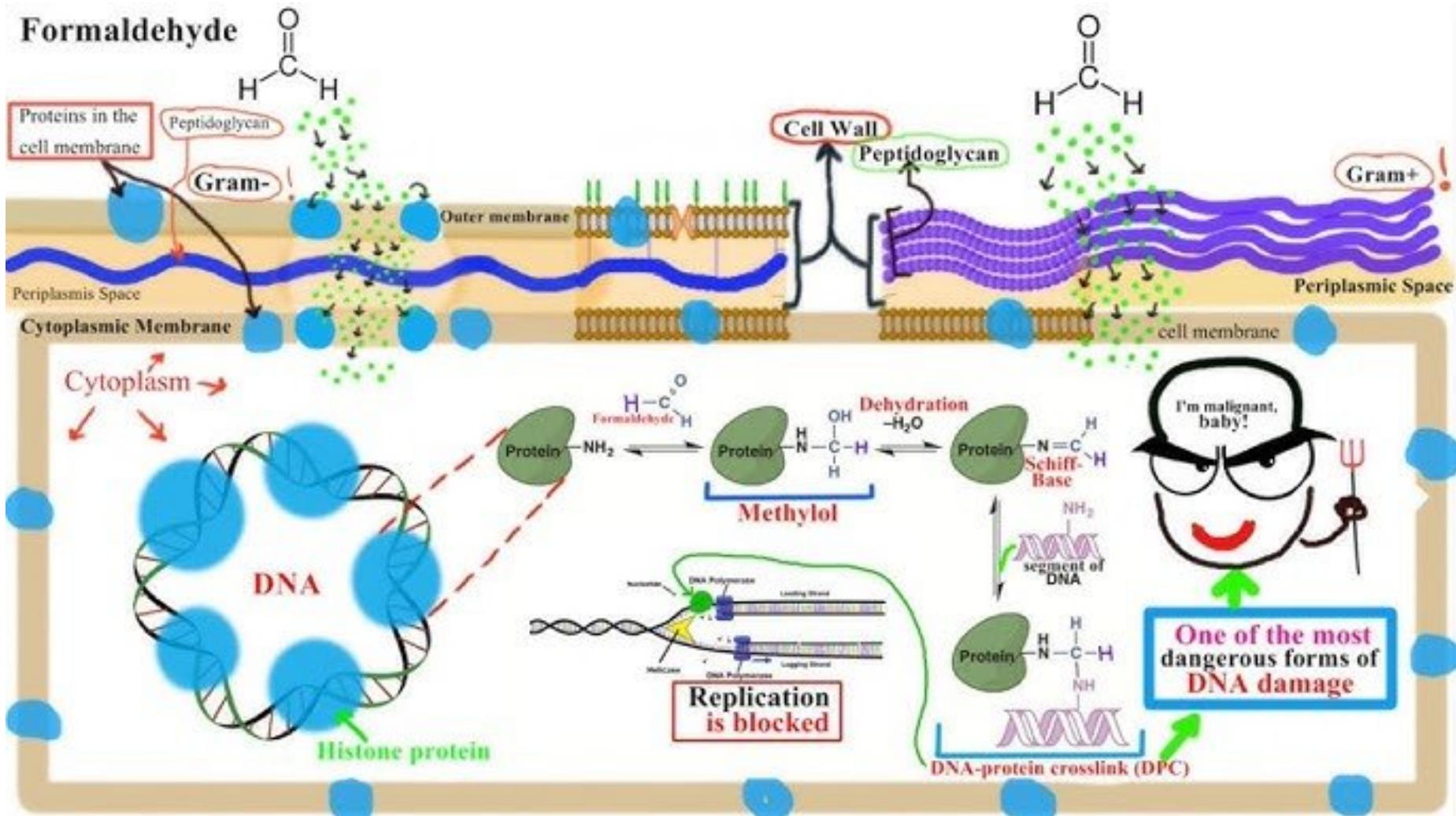
Similar action, often used in pools where it's more stable with organic matter than chlorine

Mechanism of action of Alcohols

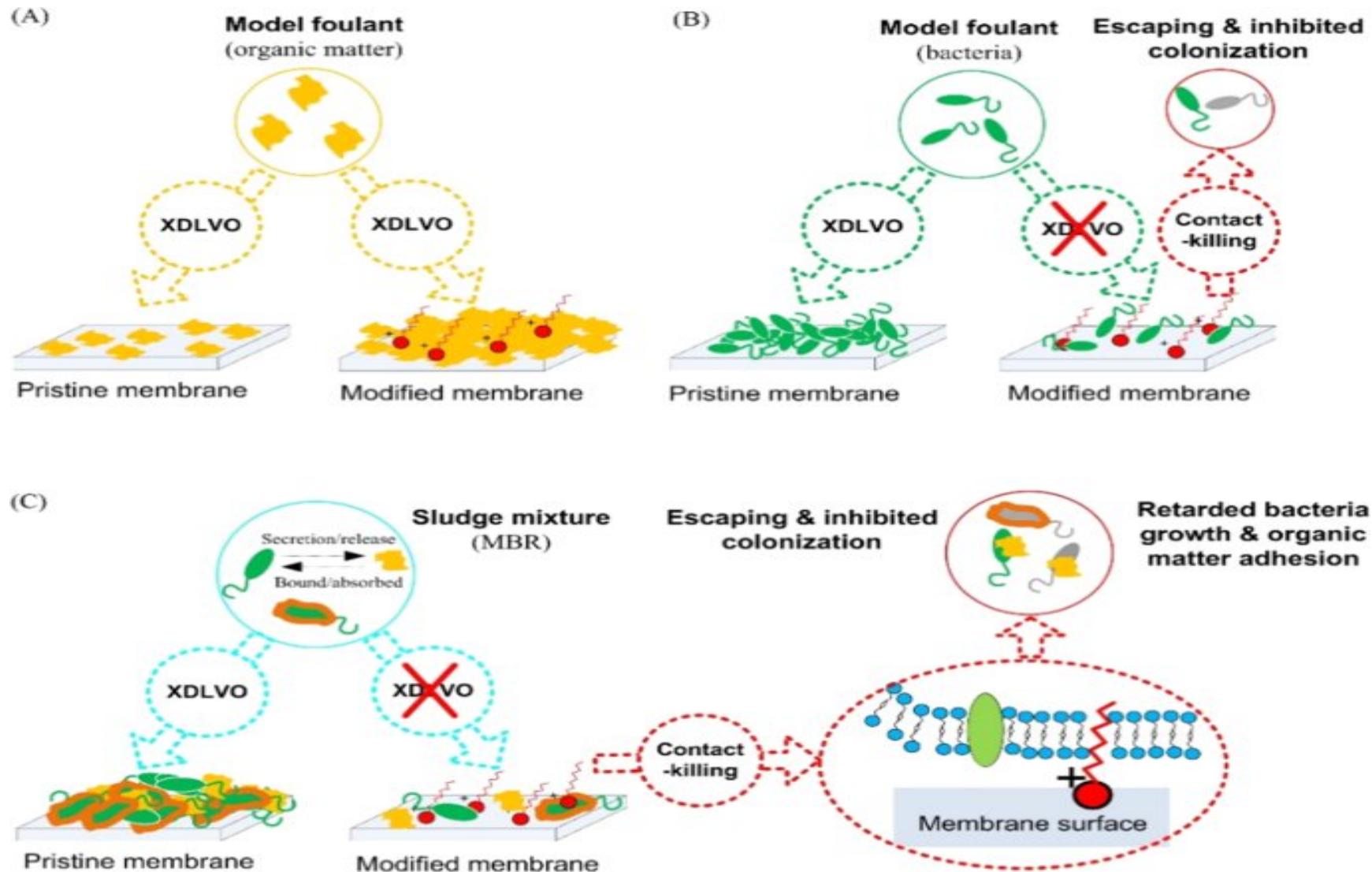


Mechanism of action of Phenolics

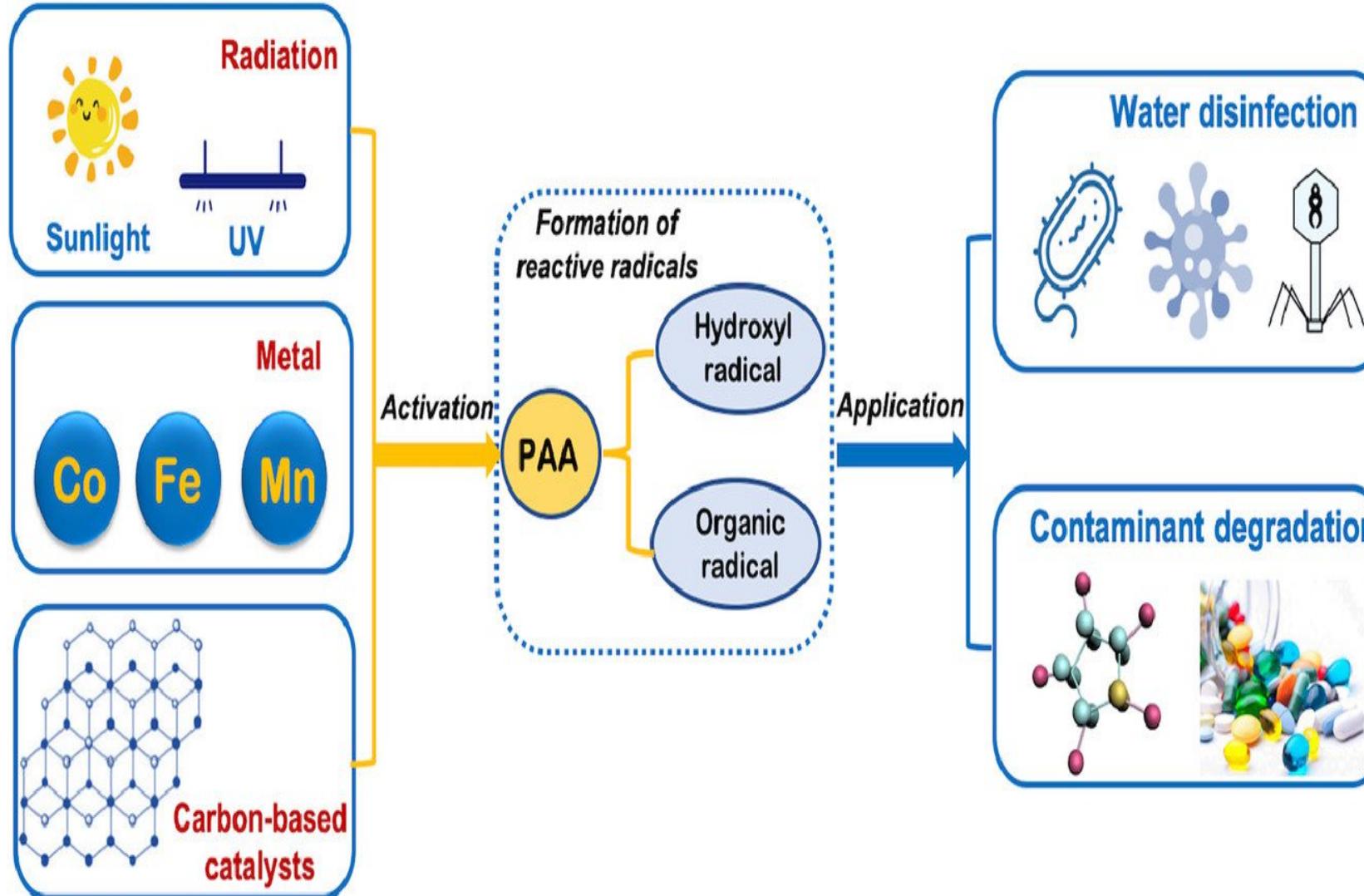




Mechanism of action of Quaternary Ammonium Compounds



Mechanism of action of Oxidizing agents





2: The Leaky Balloon Tragedy

After disinfectant exposure, the bacterial cell looks like a balloon with too many holes.

A. DNA filed a police complaint



B. Cell membrane integrity got ruined.



C. Ribosomes went on strike



D. Vitamins were confiscated



3: The Oxygen Overachiever

This disinfectant is extremely energetic and starts stealing electrons like a villain.

Which mode of action fits best?

A. Oxidative damage — too much chemistry enthusiasm 



B. Free Wi-Fi disruption 



C. Dehydrating bacteria emotionally 

D. Locking microbes in quarantine 







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2. Prescott and Dunn., Industrial Microbiology, 4th edition, CBS Publishers & Distributors, Delhi.
3. Ananthanarayan : Text Book of Microbiology, Orient-Longman, Chennai

