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ELECTROPLATING:



- Electroplating is the process in which the coating metal is deposited on the base metal by passing a direct current through an electrolytic solution contain in the soluble salt of a coating metal. In the electrolytic cell, the base metal to be placed acts as cathode where as the coating metal or good electrical conducting inert material acts as an anode.



OBJECTIVES OF ELECTROPLATING



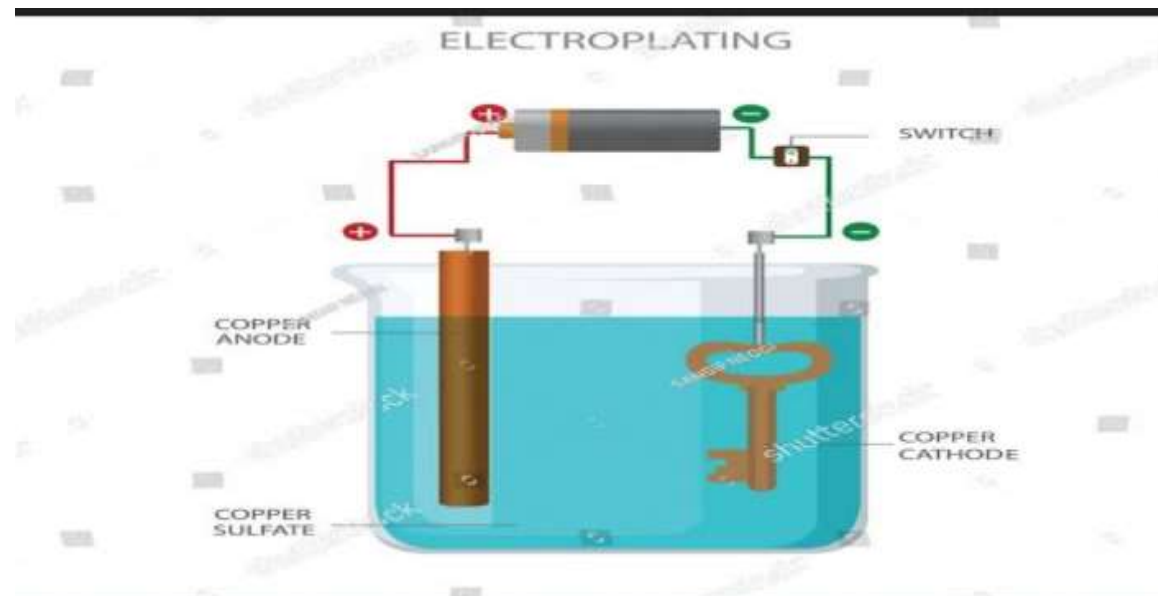
- ✓ To increase the resistance to corrosion of the base metal.
- ✓ To increase the decorative and commercial values of the articles.
- ✓ To improve the hardness and physical appearance of the articles.
- ✓ To obtain a polished surface.



PRINCIPLE:



- If the anode is made of coating metal itself in the electrolytic cell, during electrolysis, the concentration of electrolytic both remains unaltered, since the metal ions deposited from the bath on cathode are replenished continuously by the reaction of free anions with the anode.





PROCESS:



- The copper objects to be placed is first treated with dil. HCl or dil.H₂SO₄. The cleaned object is then made cathode of an electrolytic cell and gold foil as an anode. AuCl₃ solution is taken as the electrolyte. When the current is passed from the battery through the solution, gold dissolves in the electrolyte and deposits uniformly on the copper object. Thus an thin layer of gold is obtained on the article (at cathode). Sodium thiosulphate or Gelatin is used as additives.



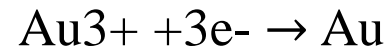
REACTIONS:



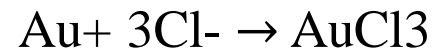
AuCl₃ ionizes as



AT CATHODE: on passing current, Au³⁺ ions move towards the cathode and get deposited on the Au object.



AT ANODE: The few Cl⁻ ions move towards the gold anode and dissolve an equivalent amount of Au to form AuCl₃.





CHARACTERISTICS OF GOLD PLATING:



- For jewelry very thin coating is given (0.05 – 1.0 microns)
- It gives high quality decorations and high oxidation resistance coating.

APPLICATIONS:

- Gold plating is used in modern semi conductor technology
- It provides corrosion protection and decoration finishes.

THANK YOU