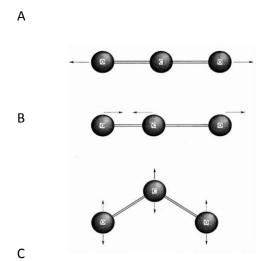
#### **Carbondioxide laser**

Carbon dioxide laser developed by Indian born American scientist Prof.C.K.N.Patel (Chandra Kumar Naranbhai Patel) at AT & T Bell laboratories,USAintheyear1963.Itisafour-levellaser,operatesat

10.6 $\mu$ minthe far IRregion. It hasaveryhighefficiencyand iswidely used for industrial drilling, cutting and welding. It is also used in medical surgery especially in cosmetic aspects. CO<sub>2</sub> laser is a molecular gas laser in which laser action is achieved by transitions betweenvibrationaland rotational levels of molecules. Its construction issimpleandtheoutputofthislaseriscontinuous.IntheCO2 moleculargas laser, transitiontakesplace betweenthe vibrational states of Carbon dioxide molecules.

#### Vibrationalenergylevelsofcarbondioxidemolecule



Carbondioxide molecule hasthreedifferent modesofvibrationshown below. The energies of associated with each of these vibrations are quantized in different sets.

**Symmetric stretching (a):** In this mode, carbon atom remains stationery when the oxygen atoms oscillate simultaneously along the molecular. In this state  $CO_2$  molecule will have energy that is intermediate as compared to that in the other two modes of vibrations (b and C). This state is referred in the spectroscopy as 100 state.

**Asymmetric stretching (b):**Herethe two oxygenatoms move in one direction while the carbon atom moves in the opposite direction and vice versa. In this stretching mode the molecule possesses highest energy and this state is represented as 001 state.

**Bending mode** (c): In bending mode, all the three atoms oscillate normal to the molecular axis. While vibrating, the two oxygen atoms pull together in one direction as the carbon atom is displaced in the opposite direction. This state is referred to a 010 state and the energy of the molecule in this state is the least among the three states of vibrations.

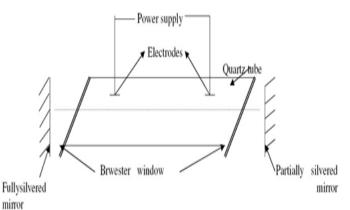
Note that 010 is the lowest excited state of bending mode. Next excited state in this mode is 020 state which is also involved in the laser transition in  $CO_2$  laser. The energy of this state

is veryclose to the 100 state of symmetric stretching mode.

#### **ConstructionofCO**<sub>2</sub>**laser:**

 $\begin{array}{c} CO_2 \text{ laser consists of a quartz tube 5 m long and 2.5 cm in the diameter. This discharge} \\ tube & is & filled & with & gaseous & mixture & of \\ CO2(activemedium), Nitrogen(N_2) and Helium(H_2) gasses in the ratio \\ \end{array}$ 

of 1:2:3 with suitable partial pressures. Sometimes traces of Hydrogen or water vapour



are added. The pressure inside the tube is 6 -17 torr.

The terminals of the discharge tubes are connected to a D.C power supply. The ends of the discharge tube are fitted with NaCl Brewster windows so that the laser light generated will be polarized. The opticallyplane mirrors fixed oneither side of the normalto its axis as showninthe figure. Oneof the mirrors is fully silvered and can reflect all the incident light whereas the other one is partially silvered to get polarized laser beam output.

## Working:

When a suitable voltage is applied across the two electrodes, a glow discharge of the gasses is initiated in the tube. During discharge the free electron from gas are accelerated towards positive electrode. Thesefree electrons collide with the  $N_2$  molecules and hence  $N_2$  molecules are raised to the first vibrational level which is meta stable

state designated as v=1 as shown in energy leveldiagram. This type of collision is called collision of first kind and can be written as

 $e_1 + N_2 = e_2 + N_2^*$ 

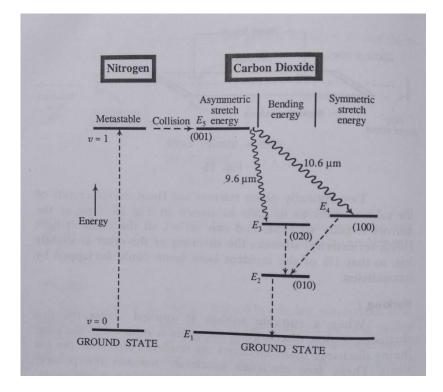
where  $e_1$  and  $e_2$  are electrons with different energies before and after

collision respectively and N<sub>2</sub> and N<sub>2</sub> \* are then it respectively.

Sincev=1stateisthe metastablestateofN<sub>2</sub>, the moleculesremainthere for a relatively long time which leads to an increase of population in v=1state. The vibrationalenergylevelcorresponding to the metastable state of N<sub>2</sub>(v=1) is close coincidence with the energy of 001 state of CO<sub>2</sub>. Because of matching of energy levels, there is a resonance transfer of energy takes place from the N<sub>2</sub> molecule to a CO<sub>2</sub> molecule through collisions. As a result, agreater number of CO<sub>2</sub> molecules get elevated to (001) state and N<sub>2</sub> molecule returns to the ground state. This type of collisions belongs to the class of collision of second kind and represented as

 $N_2^* + CO_2 = N_2 + CO_2^*$ .

Where  $CO_2$  and  $CO_2^*$  refer to the carbon dioxide molecule in the ground and excited state respectively. N<sub>2</sub><sup>\*</sup> and N<sub>2</sub> are nitrogen molecules in excited and ground states respectively



## Energy leveldiagramofCO<sub>2</sub>laser

To understand the lasing action occurring in  $CO_2$  molecule, let the ground state,(010),(020) (100),(001) are denoted as  $E_1$ ,  $E_2$ ,  $E_3$ ,  $E_4$  and  $E_5$  respectively. Due to the resonance transfer, the state  $E_5$  is heavily populated and achieves population

inversion with respect to  $E_3$  and  $E_4$ . Photons are emitted spontaneously by a few of the atoms at the energy level  $E_5$ , these spontaneous photons travelling through the gas mixture triggers lasing between  $E_5\& E_3$  and  $E_5\& E_4$  energy levels.

- a) The transition from  $E_5$  level to the  $E_4$  level gives rise to radiation of wavelength 10.6  $\mu$ m which is in the far infrared region
- b) The transition from  $E_5$  level to  $E_3$  level gives rise to radiation of wavelength 9.6  $\mu$ m which is also in the far infrared region.

## Furthertransitions

- InCO<sub>2</sub> moleculeE<sub>3</sub>-E<sub>2</sub> $\cong$  E<sub>2</sub>-E<sub>1</sub> As a consequence
- The CO<sub>2</sub> molecules in the E<sub>3</sub> level come down to E<sub>2</sub>level during collision with those in the E<sub>2</sub>level by giving up energy.
- TheCO<sub>2</sub> molecules in the E<sub>1</sub> level collide with those in E<sub>3</sub> level and are elevated E<sub>2</sub> level by absorbing the same energy.
- SinceE<sub>3</sub> ~E<sub>4</sub>thenE<sub>3</sub> −E<sub>2</sub>≅ E<sub>2</sub>−E<sub>1</sub>. This interaction is hindered by computing mechanism as follows.
- In the above energy level diagram, (E<sub>2</sub> E<sub>1</sub>) is of the order of the thermalenergyofthe surroundings. Hence excitationofCO<sub>2</sub> atoms from E<sub>1</sub> to E<sub>2</sub> takes place thus leaving less energy states in E<sub>2</sub> forE<sub>4</sub> to E<sub>2</sub> transition.
- This high population of molecules at E<sub>2</sub> level is not desirable because this will hinder the depopulation E<sub>3</sub> and E<sub>4</sub> levels which affects the population inversion of E<sub>5</sub> level.
- The collisionofmolecules in the  $E_2$ levelwithHe and water vapour leads to a quick  $E_2$ to  $E_1$ transition favouring the depopulationof $E_3$  and  $E_4$  levels and population inversion of  $E_5$  with  $E_3$  and  $E_4$  levels

The IR photons released parallel to the axis of the tube are reflected backandthemultiplication of stimulated emission takes place. The

multiple reflections of the photons between the mirrors intensify the laser beam and it emerges through the partially reflecting mirror.

The laser beam consists of two components i.e  $10.6\mu m$  and  $9.6\mu m$ .One of the components

of the beam can be eliminated by placing a

prismbetweentheBrewsterwindowandthepartiallyreflecting mirror. The laser bem passing through the prism is dispersed, the mirror is placed normaltothecomponent that isto beamplified,thiscomponent is reflected back into the laser tube and gets amplified. The other component ill not be reflected back into the tube and hence is eliminated. CO<sub>2</sub> laser operates with an efficiency of upto 30%. Power outputofa fewkilowattscanbemaintained continuouslyina medium size unit.

# Note: during discharge in the tube, some CO2 molecules breakinto CO and O. The added hydrogen or water vapour help to re oxidise CO to CO<sub>2</sub>.

## ApplicationsofLaserinMedicalfield:

Laser in eye surgery: Historically, ophthalmology was the first medical field where lasers have found an application. Over more than five decades, use of lasers in ophthalmology has successfully shown effective and safe results in treating various eye conditions. Whether lasers are used to correct vision (refractive eye surgery) or repair damage(diabeticretinopathy)duetodegenerativediseasesdictates optimal laser type, wavelength, and pulse length. Lasers are also used for cataract treatment in ophthalmology.

**Refractive eye surgery:** LASIK eye surgery is the best known and most commonly performed laser refractive surgery to correct vision problems. Laser-assisted in situ keratomileusis (LASIK) can be an alternativeto glassesorcontact lenses. Ineyes withnormalvision, the cornea bends (refracts) light preciselyontothe retina at the back ofthe eye. But with near sightedness (myopia), far sightedness (hyperopia)orastigmatism, the light isbent incorrectly, resulting in blurred vision.

**Cataract Treatment:** Cataracts, or clouding of the eye's natural lens, are the most common cause of vision loss in the elderly population. Accordingly, cataract surgery is the most commonly performed surgical procedure in the world. Treatment involves removal of thelens for a prosthetic replacement. Lasers have recently beendeveloped as a means to remove the lens via photodisruption. **Femtosecondlasers**are used in the treatment of cataracts and take advantage of the extremely high peak power densities to efficiently disrupt tissue with minimal surrounding thermal damage.

Retinal(Micro) Coagulation: Diabetic macular edema is the most commoncauseofvisualloss inpersonsunder50 yearsofage in the developed world. Retinalphotocoagulation has been developed and recognized as a standard of treatment. Recently, the procedure was perfected due to introduction of micro-coagulation technique employing pulsed **diode lasers**.